

AICE Marine Science Semester 1 Review

Unit 1 – Water

Key Vocabulary:

- | | | |
|---------------------------|--------------------------|------------------------|
| • Kinetic particle theory | • Polar | • Thermocline |
| • Water vapour | • Solvent | • PH |
| • Proton | • Density | • Acidic |
| • Neutron | • Specific heat capacity | • Alkaline |
| • Electron | • Solute | • Neutral |
| • Nucleus | • Solution | • PH scale |
| • Shells | • Solubility | • Turbulence |
| • Atomic number | • Dissolution | • Dissolved oxygen |
| • Bond | • Salinity | • Phytoplankton |
| • Emergent properties | • Precipitation | • Oxygen minimum layer |
| • Covalent bond | • Run-off | • Water column |
| • Ionic bond | • Evaporation | • Gradient |
| • Hydrogen bond | • Hypersaline | • Thermal insulator |
| • Ion | • Halocline | • Pycnocline |

1. List the 5 oceans and know their locations.
2. Identify the 4 main layers/zones of the ocean and which ones are photic?
3. Identify when you would use the different words: pelagic and benthic?
4. Explain the kinetic particle theory of WATER and the differences between solids, liquids and gases.
5. Identify the following items found in the Ocean
 - a. Key elements
 - b. Key compounds (ionic and covalent)
 - c. Key mixtures
6. Draw the structure of the following atoms (include nucleus, protons, neutrons, and electrons (show arrangement of electrons and shells/rings))
 - a. Oxygen
 - b. Hydrogen
 - c. Carbon
 - d. Sodium
 - e. Chlorine-35
7. Using a diagram describe the covalent bonding in a water molecule including the sharing of electron pairs.

Marine Science Honors Second Semester Review

Clemens Wendtner



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