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web hexane is a nonpolar liquid with a dipole moment of zero and therefore does not significantly interact with the ions of the nacl crystals 11 a fe no 3 3 is a strong electrolyte thus it should completely dissociate into fe 3 and no 3 ions therefore z best represents the solution b fe no 3 3 s fe 3 a q 3 no 3 a q

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web here the value of concentration of water is taken as 0.36 instead of 0.16 divide the product of the concentration of water carbon monoxide and the equilibrium constant by the concentration of hydrogen instead of multiplying it correct d the value of  $K_{eq}$  is taken as 4.4 instead of 0.44