

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Modern Chemistry Chapter 3 Section 3 Answers

Manfred Braun



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Principles of Organic Chemistry Peter R. S. Murray, 1977 Modern Enolate Chemistry Manfred Braun, 2016-02-23 Authored by one of the world's leading synthetic chemists in the field this reference presents modern enolate chemistry with an emphasis on metal O enolates in asymmetric synthesis While great care is taken to cover novel successful concepts such as classical methods as the famous Evans enolates are equally highlighted Throughout the book representative reaction procedures are presented thus helping readers to find the best solution for their own synthetic problem Of high interest to synthetic chemists in academia as well as the pharmaceuticals agrochemicals and fine chemicals industries *Report: Analysis of present conditions* India. Calcutta University Commission, 1919 **Educational Times**, 1909 **The Educational Times, and Journal of the College of Preceptors**, 1915 *Scribner's Monthly* Josiah Gilbert Holland, Richard Watson Gilder, 1913 **East India (Calcutta University Commission)** India Calcutta University Commission, 1919 **Decorator**, 1925 *Educational Times and Journal of the College of Preceptors*, 1908 *Report* Commonwealth Shipping Committee, 1919 *Report* Calcutta University Commission, 1919 **Report** Calcutta (India) University commission. 1917-1919, 1919 v. 4-5, Recommendations of the Commission India. Calcutta University Commission, 1917-1919, 1919 **Education Outlook**, 1914 Market World and Chronicle, 1910 **The Open Shelf**, 1915 **The A.M.A.**, 1944 *Modern Biopharmaceuticals, 4 Volume Set* Jörg Knäblein, 2005-10-28 The biopharmaceutical market has come along way since 1982 when the first biopharmaceutical product recombinant human insulin was launched Over 120 such products are currently being marketed around the world including nine blockbuster drugs The global market for biopharmaceuticals which is currently valued at US 41 billion has been growing at an impressive compound annual growth rate of 21% over the previous five years With over one third of all pipe line products in active development are biopharmaceuticals this segment is set to continue outperforming the total pharmaceutical market and could easily reach US 100 billion by the end of this decade The Athenaeum James Silk Buckingham, John Sterling, Frederick Denison Maurice, Henry Stebbing, Charles Wentworth Dilke, Thomas Kibble Hervey, William Hepworth Dixon, Norman Maccoll, Vernon Horace Rendall, John Middleton Murry, 1871

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