

# NAMING IONIC COMPOUNDS

Name Key

Name the following compounds using the Stock Naming System.

1.  $\text{CaCO}_3$  calcium carbonate
2.  $\text{KCl}$  potassium chloride
3.  $\text{FeSO}_4$  iron (II) sulfate
4.  $\text{LiBr}$  lithium bromide
5.  $\text{MgCl}_2$  magnesium chloride
6.  $\text{FeCl}_3$  iron (III) chloride
7.  $\text{Zn}_3(\text{PO}_4)_2$  zinc phosphate
8.  $\text{NH}_4\text{NO}_3$  ammonium nitrate
9.  $\text{Al}(\text{OH})_3$  aluminum hydroxide
10.  $\text{CuC}_2\text{H}_3\text{O}_2$  copper (I) acetate
11.  $\text{PbSO}_3$  lead (II) sulfite
12.  $\text{NaClO}_3$  sodium chlorate
13.  $\text{CaC}_2\text{O}_4$  calcium oxalate
14.  $\text{Fe}_2\text{O}_3$  iron (III) oxide
15.  $(\text{NH}_4)_3\text{PO}_4$  ammonium phosphate
16.  $\text{NaHSO}_4$  sodium hydrogen sulfate or sodium bisulfate
17.  $\text{Hg}_2\text{Cl}_2$  mercury (I) chloride
18.  $\text{Mg}(\text{NO}_3)_2$  magnesium nitrate
19.  $\text{CuSO}_4$  copper (II) sulfate
20.  $\text{NaHCO}_3$  sodium hydrogen carbonate or sodium bicarbonate
21.  $\text{NiBr}_3$  nickel (III) bromide
22.  $\text{Be}(\text{NO}_3)_2$  beryllium nitrate
23.  $\text{ZnSO}_4$  zinc sulfate
24.  $\text{AuCl}_3$  gold (III) chloride
25.  $\text{KMnO}_4$  potassium permanganate

baking soda  
↓ (Ammonium)

# Naming Binary Covalent Compounds Answer Key

**Martin Silberberg, Libby Weberg**



## **Naming Binary Covalent Compounds Answer Key:**

*An Introduction to Chemistry* Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life. It provides a major overview of topics typically found in first year chemistry courses in the USA. The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box. Numerous examples are presented in every chapter to aid students and provide helpful self learning tools. The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry.

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review of important concepts detailed solutions to exercises answered in the text and self test questions for each chapter

College Chemistry Steven Boone, Drew H. Wolfe, 2011-11-01 The Collins College Outline for College Chemistry is a comprehensive guide to the fundamental concepts behind chemical reactions bonding equilibria and thermodynamics with topics ranging from simple chemical measurements and the basics of atoms and molecules to entropy electrochemistry and nuclear chemistry Fully revised and updated by Dr Steven Boone College Chemistry includes practical test yourself sections with answers and complete explanations at the end of each chapter Also included are essential vocabulary definitions and sample exercises as well as detailed images charts and diagrams The Collins College Outlines are a completely revised in depth series of study guides for all areas of study including the Humanities Social Sciences Mathematics Science Language History and Business Featuring the most up to date information each book is written by a seasoned professor in the field and focuses on a simplified and general overview of the subject for college students and where appropriate Advanced Placement students Each Collins College Outline is fully integrated with the major curriculum for its subject and is a perfect supplement for any standard textbook **Chemistry** David W. Oxtoby, Norman H. Nachtrieb, Wade A. Freeman, 1994 **Chemistry for**

**the Health Sciences** George I. Sackheim, Dennis D. Lehman, 1998 This bestseller emphasizes the practical aspects of general organic and biological chemistry with numerous applications to and case histories of clinical nursing and health related situations Avoiding excessive math and theory it offers thorough and uniquely diverse coverage giving allied health professionals the chemical background necessary to understand the various medical tests and procedures they will be following and performing in their jobs Stresses the relationship between inorganic chemistry and the life processes with discussions of acids and bases oxidation reduction nuclear chemistry and radio activity and more Explains the various chemical processes taking place in the body during normal and abnormal metabolism and considers the effects of an excess or deficiency of vitamins and hormones Amazon com viewed May 14 2021 **Introductory Chemistry for Today** Spencer

L. Seager, Michael R. Slabaugh, 1999-10-22 This alternate edition is a paperback book designed for professors who want to cover only introductory chemistry or the first 12 chapters of the main text CHEMISTRY FOR TODAY GENERAL ORGANIC AND BIOCHEMISTRY Fourth Edition The ancillaries and web site that accompany the main text are also available for this briefer edition Prentice Hall Chemistry, 2000 Study Guide for Chemistry, Third Edition [by] Steven S. Zumdahl Paul B. Kelter, 1993 **General Chemistry with Qualitative Analysis** Ralph H. Petrucci, Robert K. Wismer, 1987 Study Guide, Chemical Principles, Fifth Edition, Zumdahl Paul B. Kelter, Steven S. Zumdahl, 2004-04 **Chemical Principles** Steven S. Zumdahl, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples Chemistry John S. Phillips, Cheryl Wistrom, 2000 **Chemistry: A Basic Introduction** G. Tyler Miller, Jr., 1981 *Instructor's Resource Manual* Silberberg, 1999-07

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