

Naming Polyatomic Ionic Compounds Worksheet

Name the following ions.

- 1) $(\text{NH}_4)_2\text{CrO}_4$ _____
- 2) AgNO_3 _____
- 3) $\text{Al}(\text{ClO}_3)_3$ _____
- 4) $\text{Ba}(\text{ClO}_4)_2$ _____
- 5) CaSO_4 _____
- 6) $\text{Cd}(\text{OH})_2$ _____
- 7) $\text{Cu}(\text{ClO}_4)_2$ _____
- 8) $\text{Fe}(\text{HSO}_3)_2$ _____
- 9) $\text{Fe}(\text{NO}_2)_2$ _____
- 10) $\text{Hg}_2(\text{NO}_3)_2$ _____
- 11) HgCrO_4 _____
- 12) $\text{K}_2\text{Cr}_2\text{O}_7$ _____
- 13) $\text{Mg}_3(\text{PO}_4)_2$ _____
- 14) $\text{Pb}(\text{CO}_3)_2$ _____
- 15) $\text{Pb}(\text{SO}_4)_2$ _____
- 16) $\text{Pb}_3(\text{PO}_4)_2$ _____
- 17) $\text{Sn}(\text{CN})_2$ _____
- 18) $\text{Sn}(\text{CO}_3)_2$ _____
- 19) $\text{Sr}(\text{NO}_3)_2$ _____
- 20) $\text{Zn}(\text{HCO}_3)_2$ _____

Naming Ionic Compounds With Polyatomic Ions Answers

Yicheng Fang



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