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web the heat of reaction ΔH_{soln} is written after the products in units of kJ mol^{-1} rxn for example sodium hydroxide dissolves exothermically releasing 44.2 kilojoules per mole dissolved this process is represented as $\text{NaOH (s)} \rightarrow \text{NaOH (aq)}$ $\Delta H_{\text{soln}} = -44.2 \text{ kJ mol}^{-1}$ rxn write equations to similarly represent the dissolving process for

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