

INSTRUCTIONAL PROCEDURES

4. What else could we have done on this experiment to make it more reliable?
 - Students write a conclusion about how size of surface affects the rate of dissolving sugar in water using **Solutions in a Cup Handout**.

EXPLORE/EXPLAIN

1. Bring class together for final observation.
2. Teacher pours the solution from either cup into a beaker and heats it on the hot plate.
3. Ask:
 - What do you think is going to happen?
 - Why?
4. Continue heating until all of the water is gone.
5. Students observe what is left at the bottom of the beaker.
6. Make sure to let them know that the same amount of sugar that they started with should be there at the bottom of the beaker. (Unless some of it stayed in the other cup. The main point is that the sugar did not evaporate)
7. Students explain what happened to the solution when heat was applied in **Solutions in a Cup Handout**.
8. Review vocabulary of instruction: dissolve, mixture, ingredient, substance, solution, heat resistant.

Notes for Teacher

Approximate time: Period



Materials Needed:

- Handout: **Solutions in a Cup**
- Hot plate
- beaker



Safety:
Make sure that the students don't get too close and to wear their goggles.

Teacher Review:
This would be a good time review the changes in states of matter.
Liquid → Gas (Evaporation Process)

EXPLORE

1. Review/recap the previous day activities and vocabulary.
2. Teacher models safety guidelines.
3. Introduce vocabulary of instruction.
4. Divide class into small groups and assign roles.
5. Say, "Today you will be working on an experiment. You will have to work in groups, but you will each have to turn in individual work."

Approximate time: Period



Materials Needed:

- Handout: **Dissolving Sugar**
- 3 small clear plastic cups (per group)
- 3 sugar cubes (per group)

Mid Lesson Practice Page A Tesccc Key

**Frederic Butterfield Knight, John W.
Studebaker, Giles Murrel Ruch**



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