

Chemistry Worksheets Class 11 on Chapter 11 The p-Block Elements- Set 5

Q-1: Which of the following Group 14 elements has the lowest atomic volume?

- a) Si
- b) C
- c) Sn
- d) Ge

Q-2: Which of the following is an amorphous form of carbon?

- a) Graphite
- b) Lamp-black
- c) Diamond
- d) All of the above

Q-3: Which of the following types of graphite has an ABAB-type layer arrangement?

- a) α -graphite
- b) β -graphite
- c) Both a) and b)
- d) None of the above

Q-4: Conc. HNO_3 is stored in containers made up of

- a) Zinc
- b) Aluminium
- c) Tin
- d) Copper

Q-5: Which of the following is not a protonic acid?

- a) $\text{B}(\text{OH})_3$
- b) $\text{PO}(\text{OH})_3$
- c) $\text{SO}(\text{OH})_2$
- d) $\text{SO}_2(\text{OH})_2$

Q-6: Standard electrode potential values, E° for Al^{3+}/Al is -1.66 V and $+1.26 \text{ V}$ for Ti^{2+}/Ti . Predict the creation of the M^{2+} ion in solution and contrast the electropositive properties of the two metals.

Q-7: Boron is unable to form BF_6^{3-} ion. Explain.

Q-8: Why is boric acid regarded as a weak acid?

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Survey of Science History Concepts Course Description Students will study four areas of science Scientific Mathematics Physics Biology and Chemistry Students will gain an appreciation for how each subject has affected our lives and for the people God revealed wisdom to as they sought to understand Creation Each content area is thoroughly explored giving students a good foundation in each discipline Semester 1 Math and Physics Numbers surround us Just try to make it through a day without using any It s impossible telephone numbers calendars volume settings shoe sizes speed limits weights street numbers microwave timers TV channels and the list goes on and on The many advancements and branches of mathematics were developed through the centuries as people encountered problems and relied upon math to solve them It s amazing how ten simple digits can be used in an endless number of ways to benefit man The development of these ten digits and their many uses is the fascinating story in Exploring the World of Mathematics Physics is a branch of science that many people consider to be too complicated to understand John Hudson Tiner puts this myth to rest as he explains the fascinating world of physics in a way that students can comprehend Did you know that a feather and a lump of lead will fall at the same rate in a vacuum Learn about the history of physics from Aristotle to Galileo to Isaac Newton to the latest advances Discover how the laws of motion and gravity affect everything from the normal activities of everyday life to launching rockets into space Learn about the effects of inertia first hand during fun and informative experiments Exploring the World of Physics is a great tool for student who want to have a deeper understanding of the important and interesting ways that physics affects our lives Semester 2 Biology and Chemistry The field of biology focuses on living things from the smallest microscopic protozoa to the largest mammal In this book you will read and explore the life of plants insects spiders and other arachnids life in water reptiles birds and mammals highlighting God s amazing creation You will learn about biological classification how seeds spread around the world long term storage of energy how biologists learned how the stomach digested food the plant that gave George de Mestral the idea of Velcro and so much more For most of history biologists used the visible appearance of plants or animals to classify them They grouped plants or animals with similar looking features into families Starting in the 1990 s biologists have extracted DNA and RNA from cells as a guide to how plants or animals should be grouped Like visual structures these reveal the underlying design of creation Exploring the World of Biology is a fascinating look at life from the smallest proteins and spores to the complex life systems of humans and animals Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something

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