

## THE MOLE: LAB ACTIVITY

### Station 1:

A) Determine the mass of one drop of water by measuring the mass of 50 drops of water.

Mass of dish (or beaker)	=	_____ g
Mass of dish (or beaker) + 50 drops water	=	_____ g
Mass of 50 drops water	=	_____ g
Mass of 1 drop of water	=	_____ g

B) How many molecules are there in one drop of water?

Remember 1 mole of a substance is  $6.02 \times 10^{23}$  particles (atoms or molecules).  
Therefore,  $6.02 \times 10^{23}$  molecules of water will weigh 18.02 g/mol

- Find # of moles of water for 1 drop  $n = (\text{mass})/(\text{Molar Mass})$
- # of molecules in one drop of water  $= n_{\text{water}} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole}}$   
 $= ? \text{ molecules}$

### Station 2

Calculate the number of copper atoms in a penny.

- Find # of moles of Cu  $n = (\text{mass})/(\text{Molar Mass})$
- # of Cu atoms in a penny  $= n_{\text{Cu}} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mole}}$   
 $= ? \text{ atoms}$

### Station 3

A. Calculate the number of moles in one spoon of sugar. Assume that sugar is made up entirely of sucrose molecules ( $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ ).

$$M_{\text{sucrose}} = 342.34 \text{ g/mol}$$

$$n_{\text{sugar}} = \frac{M_{\text{sucrose}}}{M_{\text{sucrose}}}$$

$$= \underline{\hspace{2cm}} \text{ mol}$$

B. Calculate the number of molecules of sugar in one spoon of sugar.

$$\# \text{ of molecules sucrose} = n_{\text{sucrose}} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}}$$

$$= \underline{\hspace{2cm}} \text{ mol} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mole}}$$

# Mole Lab Activity Answers

**R Pring**



## **Mole Lab Activity Answers:**

## The Enigmatic Realm of **Mole Lab Activity Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Mole Lab Activity Answers** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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