

Assignment #1 – Compound Names and Formulas
Single-valent ions only

A. Name these Compounds

1. Li_2S Lithium Sulfide
2. CaO Calcium Oxide
3. NaF Sodium Fluoride
4. CaBr_2 Calcium Bromide
5. MgCl_2 Magnesium Chloride
6. BBr_3 Boron Bromide
7. Cs_2O Cesium Oxide
8. FrBr Francium Bromide
9. Ag_2S Silver Sulfide

10. GeF_4 Germanium Fluoride
11. Ga_2O_3 Gallium Oxide
12. EsCl_3 Einsteinium Chloride
13. Fm_2O_3 Fermium Oxide
14. Mg_3N_2 Magnesium Nitride
15. Rb_2O Rubidium Oxide
16. RaO Radium Oxide
17. SrO Strontium Oxide
18. Tc_2O_7 Technetium Oxide

B. Write the correct chemical formula for these compounds by balancing the ionic charges

1. sodium chloride NaCl
2. magnesium fluoride MgF_2
3. silver oxide Ag_2O
4. indium bromide InBr_3
5. zinc bromide ZnBr_2
6. neodymium oxide Nd_2O_3
7. thorium sulphide ThS_2
8. actinium oxide Ac_2O_3
9. radium bromide RaBr_2
10. cesium oxide Cs_2O

11. hydrogen oxide H_2O
12. francium nitride Fr_3N
13. rubidium phosphide Rb_3P
14. potassium oxide K_2O
15. beryllium sulphide BeS
16. lithium sulphide Li_2S
17. hydrogen bromide HBr
18. strontium nitride Sr_3N_2
19. calcium oxide CaO
20. tantalum nitride Ta_3N_5

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Answers

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Formulae and Nomenclature of Binary Compounds Dennis Ord, R. W. Dent, 1964 **Inorganic Compound**

Nomenclature Evelyn Biluk, 2015-02-21 A workbook for chemistry students on naming ionic compounds binary compounds polyatomic ion compounds acids as well as chemical formulas Designed for first year undergraduate chemistry students

Free Energy of Formation of Binary Compounds Thomas Binnington Reed, 1971 Physics of Non-tetrahedrally Bonded

Binary Compounds 2 Otfried Madelung, 1983

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