

Name: _____ Date: _____ Class: _____

What do You Know About Acids and Bases?

| Letter | Matching | |
|--------|----------------|--|
| | 1. Acid | A. To mix acids and bases to cancel each other out and make water and salt |
| | 2. Base | B. A compound that adds H^+ ions to water |
| | 3. Neutral | C. Equal number of H^+ and OH^- ions; water is an example |
| | 4. Neutralize | D. A compound that adds OH^- ions to water. |
| | 5. Acid Rain | E. When pollution causes rain to have a pH less than 5.6. |
| | 6. pH | F. The measure of how acidic or basic a solution is |
| | 7. Salt Water | G. A compound that adds a few OH^- ions to water |
| | 8. Strong Acid | H. The product of a neutralization reaction between an acid and a base |
| | 9. Weak Base | I. A compound that adds a few H^+ ions to water |
| | 10. Weak Acid | J. A compound that adds a lot of H^+ ions to water. |

| Is it an Acid or a Base? | | |
|--|----------------|-----------------------|
| Circle the acids and underline the bases | | |
| 11. HCl | 17. H_2CO_3 | 23. apple juice |
| 12. $Mg(OH)_2$ | 18. NaOH | 24. lemonade |
| 13. H_3PO_4 | 19. $Al(OH)_3$ | 25. soap |
| 14. KOH | 20. HBr | 26. laundry detergent |
| 15. $Ca(OH)_2$ | 21. H_2SO_4 | 27. soft drinks |
| 16. LiOH | 22. H_2O | 28. bathroom cleaner |

Lecture 29 Acids Base I Worksheet

H Kauffman



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