



Turning the Unit ON / OFF

1. Press **SOURCE** to turn the unit ON.
2. Press and hold **SOURCE** to turn the unit OFF.

Selecting a Source

1. Press **SOURCE** to access the **SOURCE: Select** menu.
2. Rotate the **VOL/SEL** knob to highlight a desired source; press to select.

Adjusting Volume

The **VOL/SEL** knob is designed to be used as a main volume controller, allowing easy adjustment of the audio levels, for all zones simultaneously or for each zone independently. Refer to **Audio Zone Setup** on pages 15-16 for more info.

1. Rotate the **VOL/SEL** knob clockwise/counter-clockwise to increase/decrease volume (this also displays the **Audio Zone Level** screen).

Muting or Pausing Audio

1. Press **▶ ||** to mute/unmute audio (AM/FM) or pause/resume the current selection (USB/Bluetooth®).

Jl Tr60cxi User Guide

L Cohen



Jl Tr60cxi User Guide:

Whispering the Strategies of Language: An Psychological Quest through **Jl Tr60cxi User Guide**

In a digitally-driven world wherever screens reign great and quick conversation drowns out the subtleties of language, the profound strategies and mental subtleties concealed within phrases frequently move unheard. However, located within the pages of **Jl Tr60cxi User Guide** a interesting fictional value blinking with organic emotions, lies a fantastic journey waiting to be undertaken. Published by an experienced wordsmith, that wonderful opus invites readers on an introspective trip, delicately unraveling the veiled truths and profound impact resonating within the material of each and every word. Within the psychological depths of this moving review, we shall embark upon a sincere exploration of the book is primary themes, dissect its fascinating publishing style, and fail to the strong resonance it evokes deep within the recesses of readers hearts.

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 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta H = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow 2 \text{ C}(\text{s}) + 3 \text{ H}_2(\text{g}) \quad \Delta H = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1411 \text{ kJ/mol}$
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1560 \text{ kJ/mol}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta H = 285 \text{ kJ/mol}$

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