

## WORKSHEET – Naming Compounds KEY

### Part I – Naming Compounds and Numbering Atoms

FORMULA	NAME	# of DIFFERENT ATOMS	TOTAL # OF ATOMS
1. $\text{CaCl}_2$	Calcium chloride	1 Ca; 2 Cl	3
2. $\text{AgCl}$	<i>Silver chloride</i>	1 Ag; 1 Cl	2
3. $\text{MgO}$	<i>Magnesium oxide</i>	1 Mg; 1 O	2
4. $\text{NaBr}$	<i>Sodium bromide</i>	1 Na; 1 Br	2
5. $\text{Al}_2\text{O}_3$	<i>Aluminum oxide</i>	2 Al; 3 O	5
6. $\text{KI}$	<i>Potassium iodide</i>	1 K; 1 I	2
7. $\text{ZnCl}_2$	<i>Zinc chloride</i>	1 Zn; 2 Cl	3
8. $\text{Li}_2\text{O}$	<i>Lithium oxide</i>	2 Li; 1 O	3
9. $\text{BaO}$	<i>Barium oxide</i>	1 Ba; 1 O	2
10. $\text{CaBr}_2$	<i>Calcium bromide</i>	1 Ca; 2 Br	3
11. $\text{MgCl}_2$	<i>Magnesium chloride</i>	1 Mg; 2 Cl	3
12. $\text{AgI}$	<i>Silver iodide</i>	1 Ag; 1 I	2
13. $\text{ZnS}$	<i>Zinc sulphide</i>	1 Zn; 1 S	2
14. $\text{BaF}_2$	<i>Barium fluoride</i>	1 Ba; 2 F	3
15. $\text{Ca}_3\text{P}_2$	<i>Calcium phosphide</i>	3 Ca; 2 P	5
16. $\text{Na}_2\text{O}$	<i>Sodium oxide</i>	2 Na; 1 O	3
17. $\text{AlN}$	<i>Aluminum nitride</i>	1 Al; 1 N	2
18. $\text{K}_3\text{N}$	<i>Potassium nitride</i>	3 K; 1 N	4
19. $\text{Ca}_2\text{C}$	<i>Calcium carbide</i>	2 Ca; 1 C	3
20. $\text{K}_2\text{O}$	<i>Potassium oxide</i>	2 K; 1 O	3

# Naming Compounds Answer Key

**Patrick Vollmar**



## **Naming Compounds Answer Key:**

*Physical and Chemical Changes (ENHANCED eBook)* Edward P. Ortleb, Richard Cadice, 1993-09-01 This book presents a program of basic studies in physical and chemical changes of matter The definition of matter is presented along with explanations of states and properties of matter Topics include atoms molecules elements compounds mixtures solutions symbols and formulas Each of the twelve teaching units in this book is introduced by a color transparency print books or PowerPoint slide eBooks that emphasizes the basic concept of the unit and presents questions for discussion Reproducible student pages provide reinforcement and follow up activities The teaching guide offers descriptions of the basic concepts to be presented background information suggestions for enrichment activities and a complete answer key Chemistry for the Utterly Confused John T. Moore, Richard H. Langley, 2007-06-01 Banish bafflement in this tough subject From formulas and lab techniques to the periodic table Chemistry for the Utterly Confused focuses on the areas of maximum confusion and breaks down the most difficult chemistry topics into easy to understand concepts This invaluable guide also teaches problem solving skills you need to master this imposing subject Whether you re in high school in college or simply brushing up on chemistry knowledge this fun easily accessible book will make understanding chemistry a breeze **Physical Science**, 2015-03-16 Physical Science for grades 5 to 12 is designed to aid in the review and practice of physical science topics Physical Science covers topics such as scientific measurement force and energy matter atoms and elements magnetism and electricity The book includes realistic diagrams and engaging activities to support practice in all areas of physical science The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series is aligned to current science standards *EBOOK: GENERAL CHEMISTRY, THE ESSENTIAL CONCEPTS* CHANG, 2013-01-07 EBOOK GENERAL CHEMISTRY THE ESSENTIAL CONCEPTS *Chemistry Super Review* Editors of REA, 2012-05 Get all you need to know with Super Reviews Each Super Review is packed with in depth student friendly topic reviews that fully explain everything about the subject The Chemistry Super Review includes an overview of stoichiometry atomic structure and the periodic table bonding chemical formulas types and rates of chemical reactions gases liquids solids phase changes properties of solutions acids bases chemical equilibrium chemical thermodynamics oxidation and reduction Take the Super Review quizzes to see how much you ve learned and where you need more study Makes an excellent study aid and textbook companion Great for self study DETAILS From cover to cover each in depth topic review is easy to follow and easy to grasp Perfect when preparing for homework quizzes and exams Review questions after each topic that highlight and reinforce key areas and concepts Student friendly language for easy reading and comprehension Includes quizzes that test your understanding of the subject Basic Chemistry Steven S.

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