

1. How many atoms of hydrogen are in 0.500 mol of CH_3OH molecules?

A. 1.20×10^{23}
 B. 3.01×10^{23}
 C. 6.02×10^{23}
 D. 1.20×10^{24}

2. Calcium carbonate reacts with hydrochloric acid according to the following equation.



What is the theoretical yield, in mol, of calcium chloride if 0.10 mol CaCO_3 is added to 100 cm^3 of 1.0 mol dm^{-3} HCl?

A. 0.050
 B. 0.10
 C. 0.20
 D. 0.50

3. A fixed mass of an ideal gas at 27.0 °C and 1.01×10^5 Pa has a volume of 100 cm^3 . Which change doubles the volume of the gas?

A. Heating the gas at constant pressure to 54.0 °C.
 B. Heating the gas at constant pressure to 327 °C.
 C. Increasing the pressure on the gas to 2.02×10^5 Pa at constant temperature.
 D. Heating the gas to 54.0 °C and increasing the pressure to 2.02×10^5 Pa.

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