

Name : _____

Naming Acids and Bases

1) How can you tell by looking at a chemical formula that it is an acid?

2) What is an oxyacid?

Below are the formulas of several compounds. Circle the binary acids and put a square around the oxyacids

HCl H₂S HMnO₄ NaOH HC₂H₃O₂ HF Sr(OH)₂ H₂SO₄ NH₃

3) Fill in the chart about naming binary acids.

Acid Formula	Positive Ion	Negative Ion	Acid Name
Example: HF	H ⁺	F ⁻	Hydrofluoric acid
HCl			
HBr			
HI			
H ₂ S			
H ₂ Se			

4) Fill in the chart about naming oxyacids.

Acid Formula	Positive Ion	Negative Ion	Acid Name
Example: H ₂ SO ₄	H ⁺	SO ₄ ²⁻	Sulfuric acid
HMnO ₄			
HCIO ₄			
HIO ₄			
HCIO			
C ₂ H ₃ O ₄			

Naming Acids And Bases Evan P

LL Leslie



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