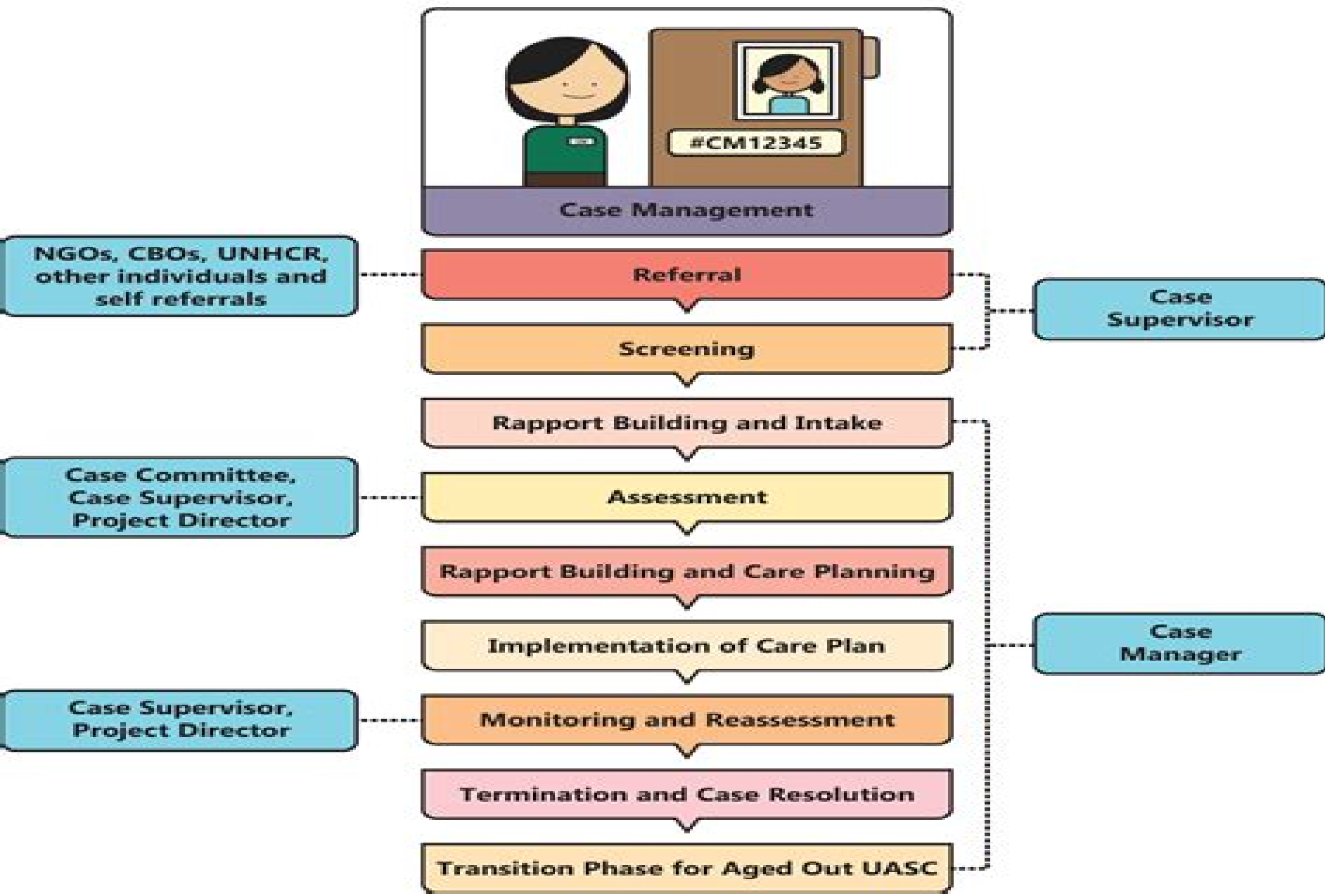


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since it would yield the least amount of product 18 mol Br_2 the excess reactant is $\text{C}_2\text{H}_3\text{Br}_3$ since its complete reaction would have yielded up to 54 mol Br_2 think about your result

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