

# PHYSICAL SCIENCE - SYLLABUS

## 10th CLASS

### 1. Reflection of light at curved surface

- 1.1 Normal to the curved surface
- 1.2 Spherical mirrors, convex, concave mirrors
- 1.3 Pole, Focus, Centre of curvature, principle axis, Radius of curvature, Focal length
- 1.4 Images formed by spherical mirrors
- 1.5 Ray diagrams for spherical mirrors
  - 1.5.1 Rules for Ray diagrams by using laws of reflection
- 1.6 Formula for spherical mirrors – sign convention
  - 1.6.1 Magnification
- 1.7 Application of reflection - Solar Cooker

### 2. Chemical Equations and Reactions

- 2.1 Some daily life examples of chemical reactions.
- 2.2 Chemical equations – writing chemical equations, skeletal chemical equations, balancing chemical equations
- 2.3 Writing symbols of physical states, Heat changes, gas evolved and precipitate formed
- 2.4 Interpreting a balanced chemical equation
  - 2.4.1 Calculations based on mass, volume, number of molecules and moles

### 3. Acids, Bases and Salts

- 3.1 Chemical properties of acids & bases

3.1.1 Acids & Bases in laboratory – Indicators

3.1.2 Reaction of Acids & Bases with Metals

3.1.3 Reaction of Acids & Bases with Metal Carbonates and Metal hydrogen carbonates

3.1.4 Reaction of Acids & Bases with each other (Neutralization)

3.1.5 Reaction of Acids with Metallic oxides

3.1.6 Reaction of Bases with Non-Metallic oxides

3.2 What do acids have in common? What do bases have in common?

3.3 Do Acids produce ions only in Aqueous Solution ?

3.4 Reaction of Acid, Base with water

3.5 Strength of Acid or Base -  $p^H$  scale

3.6 Importance of  $p^H$  in everyday life

3.6.1 Sensitivity of plants and animals to  $p^H$

3.6.2  $p^H$  of soils,  $p^H$  in digestive system,  $p^H$  tooth decay

3.6.3 Self defense by animals and plants through chemical warfare

3.7 Salts

3.7.1 Family of salts

3.7.2  $p^H$  of salts

3.8 Chemicals from common salt

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