
MICROBIOLOGY EXAM #1 STUDY GUIDE WITH COMPLETE SOLUTIONS 100%

Spontaneous Generation - correct answer The hypothesis that living organisms arise from nonliving matter. A "vital force" forms life.

Biogenesis - correct answer The hypothesis that the living organisms arise from preexisting life.

Leeuwenhoeke - correct answer Developed the first microscope.

Pasteur - correct answer Showed that microbes are responsible for fermentation. Demonstrated that these spoilage bacteria could be killed by heat.

Koch - correct answer Proved that a bacterium causes anthrax. Developed Koch's Postulates.

Koch's Postulates - correct answer The microorganism must be present in all cases of the disease.

The microorganism can be isolated from the diseased host and grown in a laboratory setting.

The microorganism will cause disease when injected into a healthy host.

The microorganism must be reisolated from the NEW host and shown to be the same as the microorganism previously identified.

Jenner - correct answer First vaccination.

Erlch - correct answer Founding father of chemotherapy

Fleming - correct answer Discovered first antibiotic

Microbiology Exam Study Guide

Jianjun Gao



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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^- + \text{Au} \rightarrow \text{Au} + \text{CN}^- + 2\text{e}^-$ $2\text{H}^+ + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$ 2 make electrons equal convert to basic solution $4\text{CN}^- + 2\text{Au} \rightarrow 2\text{Au} + \text{CN}^- + 2\text{e}^-$ multiplied by a factor of 2 $2\text{e}^- + 2\text{H}_2\text{O} \rightarrow \text{O}_2 + 2\text{OH}^-$ 3 the final answer

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web aug 29 2023 the first step to balance any redox reaction is to separate the reaction into half reactions the substance being reduced will have electrons as reactants and the oxidized substance will have electrons as products usually all reactions are written as reduction reactions in half reaction tables

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$\text{C}_6\text{H}_8\text{O}_6 + 2\text{NO} \rightarrow \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}_2$

choose 1 answer

☐ $\text{C}_6\text{H}_8\text{O}_6$ and NO_2

☐ $\text{C}_6\text{H}_8\text{O}_6$ and NO

☐ $\text{C}_6\text{H}_8\text{O}_6$ and H_2O

☐ $\text{C}_6\text{H}_8\text{O}_6$ and H_2O