

You must show all your calculation set-ups to receive full credit. You must report all answers to the correct # of significant figures (sig. figs.).

1. Who used a cathode ray tube in his/her discovery of the electron?

- ☒ a. Robert Millikan  
☐ b. Ernest Rutherford  
☐ c. John Dalton  
☒ d. J.J. Thomson

82  
100

2. How many significant figures are in each of the following measurements?

- a. 0.0017 g 2 sig figs  
 b. 1390 mL 3 sig figs  
 c. 4.030 km 4 sig figs  
 d. 200110 kg 5 sig figs

3. Which unit represents a microgram?

- ☒ a. mg  
☒ b.  $\mu$ g  
☐ c. ng  
☐ d. cg

4. The most unreactive group in the Periodic Table is/are the

- ☒ a. alkaline-earth metals  
☐ b. metalloids  
☐ c. halogens  
☒ d. noble gases

5. The empirical formula of a compound of carbon and hydrogen is  $\text{CH}_2$ . The molar mass is 28.06 g/mol. What is the molecular formula?

- ☒ a.  $\text{C}_2\text{H}_4$   
☐ b.  $\text{C}_3\text{H}_6$   
☐ c.  $\text{C}_4\text{H}_8$   
☐ d.  $\text{C}_5\text{H}_{10}$

C: 12.01  
 H: 1.01

$$28.06 = n(14.03)$$

$$n = 2$$

$$\frac{C \ 12.01}{H \ 2(1.01)}$$

$$14.03$$

6. Label each of the following elements as metal, nonmetal, or metalloid.

- a. chlorine nonmetal  
 b. beryllium metal  
 c. silicon metalloid  
 d. boron metalloid

7. Write both the name and the symbol of the element in group 13 (3A) and period 3.

Aluminum (Al)

8. Calculate the percentage composition by mass (mass percent) of just the hydrogen in  $\text{C}_3\text{H}_8$ .

$$\text{C}_3\text{H}_8 \rightarrow \text{total: } 44.11$$

$$C: 3(12.01) = 36.03 / 44.11 = 82\%$$

$$H: 8(1.01) = 8.08 / 44.11 = 18\%$$

# Michigan Standardized College Chemistry Exam

**VM Jensen**



**Michigan Standardized College Chemistry Exam:**

## The Enigmatic Realm of **Michigan Standardized College Chemistry Exam**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Michigan Standardized College Chemistry Exam** a literary masterpiece penned with a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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