

DE-100A: STUDY GUIDE

Introduction and Overview - what topics will you be learning
 - understanding basic thermodynamics + 1st, 2nd, 3rd Laws of Thermodynamics
 - **Energy**
 - **Enthalpy**
 - **Entropy**
 - **Free Energy**
 - **Equilibrium**
 - **Chemical Equilibrium**
 - **Le Chatelier's Principle**

Thermodynamics
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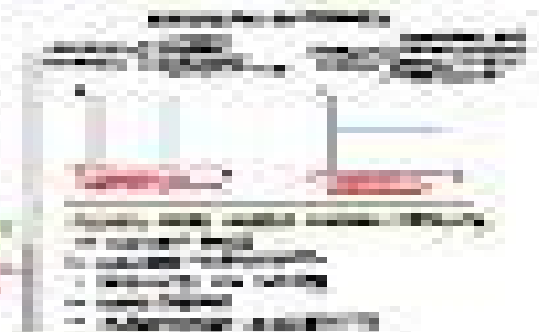
Thermodynamics - study of energy and its transformations
 - **1st Law of Thermodynamics** - Energy cannot be created or destroyed, only transferred or converted from one form to another.
 - **2nd Law of Thermodynamics** - The total entropy of an isolated system can never decrease over time, and it increases for all spontaneous processes.
 - **3rd Law of Thermodynamics** - The entropy of a perfect crystal is zero at absolute zero temperature.



Enthalpy - a thermodynamic property that represents the total heat content of a system.
 - **Enthalpy Change (ΔH)** - The difference in enthalpy between the products and reactants of a chemical reaction.
 - **Exothermic Process** - A process that releases heat to the surroundings, resulting in a negative ΔH .
 - **Endothermic Process** - A process that absorbs heat from the surroundings, resulting in a positive ΔH .



Free Energy - a thermodynamic property that represents the maximum work that can be done by a system at constant temperature and pressure.
 - **Gibbs Free Energy (ΔG)** - The difference in Gibbs free energy between the products and reactants of a chemical reaction.
 - **Spontaneous Process** - A process that occurs without the need for external energy input, resulting in a negative ΔG .
 - **Non-spontaneous Process** - A process that requires external energy input to occur, resulting in a positive ΔG .



Chemical Equilibrium - a state in which the concentrations of reactants and products remain constant over time.
 - **Equilibrium Constant (K)** - A measure of the extent to which a chemical reaction proceeds towards the products.
 - **Le Chatelier's Principle** - A principle that states that if a system at equilibrium is subjected to a change in concentration, temperature, or pressure, the system will shift in a direction that counteracts the change.

Thermodynamics - study of energy and its transformations
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Microeconomics Study Guide Final

Rosina Ehmann



Microeconomics Study Guide Final:

Whispering the Strategies of Language: An Emotional Journey through **Microeconomics Study Guide Final**

In a digitally-driven world where monitors reign great and instant conversation drowns out the subtleties of language, the profound strategies and emotional nuances concealed within words usually go unheard. Yet, nestled within the pages of **Microeconomics Study Guide Final** a interesting fictional value blinking with organic feelings, lies an extraordinary journey waiting to be undertaken. Penned by an experienced wordsmith, this enchanting opus invites viewers on an introspective trip, lightly unraveling the veiled truths and profound affect resonating within ab muscles fabric of each word. Within the psychological depths of the poignant evaluation, we shall embark upon a sincere exploration of the book is key subjects, dissect its interesting writing style, and yield to the powerful resonance it evokes serious within the recesses of readers hearts.

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