

## **SMT SCHEME OF WORK AND JAMB SYLLABUS FOR 2025 JAMBITES.**

CHEMISTRY	PHYSICS	BIOLOGY	MATHEMATICS
<ul style="list-style-type: none"> <li>➤ Separation and mixture</li> <li>➤ Chemical combination</li> <li>➤ Kinetic theory</li> <li>➤ Atomic structure and bonding</li> <li>➤ Air</li> <li>➤ Water</li> <li>➤ Solubility</li> <li>➤ Environmental pollution</li> <li>➤ Acids bases and salts</li> <li>➤ Oxidation and reduction</li> <li>➤ Electrolysis.</li> <li>➤ Organic chemistry</li> <li>➤ Rates of chemical reactions</li> <li>➤ Chemical equilibrium</li> <li>➤ Non-metals and their compounds.</li> </ul>	<ul style="list-style-type: none"> <li>➤ Measurements</li> <li>➤ Motion</li> <li>➤ Newton's laws of motion</li> <li>➤ Energy</li> <li>➤ Waves</li> <li>➤ Optics.</li> <li>➤ Electricity</li> <li>➤ Magnetism</li> <li>➤ Electronics</li> <li>➤ Thermal physics</li> <li>➤ Simple harmonic motion and waves</li> <li>➤ Nuclear physics</li> <li>➤ Electronics and telecommunication</li> </ul>	<ul style="list-style-type: none"> <li>➤ Living organisms and their interactions</li> <li>➤ Cell biology</li> <li>➤ Genetics</li> <li>➤ Evolution</li> <li>➤ Plant biology</li> <li>➤ Animal biology</li> <li>➤ Ecology</li> <li>➤ Taxonomy and diversity</li> <li>➤ Health education and human disease.</li> </ul>	<ul style="list-style-type: none"> <li>➤ Numbers and numeration</li> <li>➤ Indices</li> <li>➤ Surds</li> <li>➤ Logarithm</li> <li>➤ Sets</li> <li>➤ Inequalities.</li> <li>➤ Quadratic equation</li> <li>➤ Series and sequence</li> <li>➤ Variation</li> <li>➤ Matrices and determinants.</li> <li>➤ Angles and polygons</li> <li>➤ Geometry and trigonometry</li> <li>➤ Probability</li> <li>➤ Statistics</li> </ul>

# Jamb Scheme Of Work For Crk

**Jiyuan Zhang**



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 kg the molality equation is a mathematical formula that describes how much a substance weighs in this equation m moles  
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 with 1200 ml of 2m hno3 if the final density of solution is same as pure water then molarity of the iron in resulting solution  
 which decides the nature of the above solution is 1 0 25m 2 0 5m 3 0 888m 4 none of

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 concentration units and they are used interchangeably

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 0 10 0 0585 correct answer b 1 explanation molality begin array l m left frac 1000 m 1000d mm right end array

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 of solvent whereas molarity is measured in liters of solution when studying the properties of solutions related to vapour  
 pressure and temperature changes concentrations expressed in molality are used

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 the molar concentration of the solute is sometimes abbreviated by putting square brackets around the chemical formula of

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 signifies solutions labelled with molar concentration a 1 0 m solution contains 1 mole of solvent per litre of solution molality

is the number of solvent moles per kilogram

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web 1 molarity mv mass molar mass x 0 4500 l 0 825 g 141 9579 g mol x 0 0129 m 2 molality 0 825 g 141 9579 g mol 0 00581158 mol 0 00581158 mol 0 4500 kg 0 0129 m 3 mole fraction na 2 hpo 4 0 825 g 141 9579 g mol 0 00581158 mol h 2 o 450 0 g 18 015 g mol 24 97918401 mol

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$$m_{mix} = \frac{m_1}{V_1} = \frac{m_2}{V_2} = \frac{m_3}{V_3} = \frac{m_4}{V_4} = \frac{m_5}{V_5}$$
 1 5 x 480 1 2 x 520 480 520 720 624 1000 1 344 m q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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