#### Significant Figure and Scientific Notation Worksheet

#### 1. Convert each of the following into scientific notation.

a) 3427

ь) 0.00456

c) 123, 453

d) 172

e) 0.000984

0.502

g)  $3100.0 \times 10^{2}$ 

b) 0.0114×104

i) 107.2

j) 0.0000455

k) 2205.2

1)  $30.0 \times 10^{-2}$ 

m)  $0.982 \times 10^{-3}$ 

n) 0.0473

0) 650,502

p)  $3.03 \times 10^{-1}$ 

 $q)20.4 \times 10^{5}$ 

r) 1000×10<sup>-3</sup>

#### 2. Determine the number of significant digits in each of the following:

a) 3427

b) 0.00456

c) 123,453

d) 172

c) 0.000984

0.502

g) 3100.0×10<sup>2</sup>

b) 0.0114×104

i) 107.2

j) 0.0000455

k) 2205.2

1)  $30.0 \times 10^{-2}$ 

m)  $0.982 \times 10^{-3}$ 

n) 0.0473

o) 650,502

p)  $3.03 \times 10^{-1}$ 

q)  $20.4 \times 10^5$ 

r) 1000×10<sup>-3</sup>

#### 3. Convert each into decimal form.

a)  $1.56 \times 10^4$ 

b)  $0.56 \times 10^{-2}$ 

c)  $3.69 \times 10^{-2}$ 

d) 736.9×10<sup>5</sup>

e) 0.00259×10<sup>5</sup>

f)  $13.69 \times 10^{-2}$ 

g)  $6.9 \times 10^4$ 

# **Mastering Chemistry Answer Key Significant Figures**

**D** Kirk

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