

**CHEM 101 Quiz #6 Chapter 6 ANSWERS****1. (4 points)**

The specific heat (s) of iron (Fe) is  $0.40 \text{ J/(g K)}$ . Calculate the heat effect  $q$  when 20 g of iron is heated from  $100^\circ\text{C}$  to  $200^\circ\text{C}$ . Include the sign!

$$q = m \cdot s \cdot \Delta T \text{ for } \Delta T \Delta K = \Delta^\circ\text{C} \quad q = 20 \times 0.40 \times (200 - 100) = +800 \text{ J}$$

You must include the units and sign in the answer!

**2. (4 points)**

The Molar Mass of Mg =  $24.0 \text{ g/mol}$ .

Calculate the enthalpy change when 2.40 g Mg(s) reacts with excess oxygen. Include the sign!

$$2.40 \text{ g Mg} = 2.40/24.0 = 0.1 \text{ mol} \rightarrow \Delta H = 0.1(\text{mol}) \times (-600)(\text{kJ/mol}) = -60 \text{ kJ}$$

Again, your answer must include both the sign and the units.

**3. (2 points)** Which one of the following four reactions is a *formation reaction*? Circle the correct answer.



*Look up the definition of a formation reaction in Chang or in your lecture notes: the compound is formed from its elements, and the elements must be as they naturally occur so e.g.  $\text{Cl}_2$  not  $\text{Cl}$ , and  $\text{H}_2$  not  $\text{H}$ !*

**CHEM 101 Section L52 Quiz #6****1. (4 points)**

The specific heat (s) of copper (Cu) is  $0.38 \text{ J/(g K)}$ . Calculate the heat effect  $q$  when 10 g of copper is heated from  $100^\circ\text{C}$  to  $300^\circ\text{C}$ . Include the sign!

$$q = m \cdot s \cdot \Delta T \text{ for } \Delta T \Delta K = \Delta^\circ\text{C} \quad q = 10 \times 0.38 \times (300 - 100) = +760 \text{ J}$$

You must include the units and sign in the answer!

**2. (4 points)**

The Molar Mass of Ca =  $40.0 \text{ g/mol}$ .

Calculate the enthalpy change  $\Delta H$  when 8.0 g Ca(s) reacts with excess oxygen. Include the sign!!

$$8.0 \text{ g Ca} = 8.0/40 = 0.2 \text{ mol} \rightarrow \Delta H = 0.2(\text{mol}) \times (-650)(\text{kJ/mol}) = -130 \text{ kJ}$$

Again, your answer must include both the sign and the units.

**3. (2 points)** Which one of the following reactions is a *formation reaction*? Circle the correct answer.



# Mastering Chemistry Answers Ch 6

**Clifford Lane**



## **Mastering Chemistry Answers Ch 6:**

## Embracing the Song of Appearance: An Psychological Symphony within **Mastering Chemistry Answers Ch 6**

In a global consumed by screens and the ceaseless chatter of fast connection, the melodic elegance and mental symphony created by the written word usually disappear into the background, eclipsed by the constant noise and interruptions that permeate our lives. Nevertheless, set within the pages of **Mastering Chemistry Answers Ch 6** a marvelous fictional treasure brimming with natural emotions, lies an immersive symphony waiting to be embraced. Crafted by an outstanding composer of language, this captivating masterpiece conducts readers on a psychological journey, well unraveling the hidden songs and profound influence resonating within each carefully crafted phrase. Within the depths with this emotional review, we shall discover the book is central harmonies, analyze their enthralling publishing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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