

Name: _____

Block: _____

Date: _____

Chemistry 11

Classifying Matter

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements
Mixtures
Compound
Symbols
Properties

Carbon
Pure
Letters
Carbon dioxide

- a. Elements cannot be separated by physical or chemical means.
- b. Elements are abbreviated with symbols, which consist of one or two letters.
- c. Two or more elements chemically combined make up a compound.
- d. Carbon is an example of an element and carbon dioxide is an example of a compound.
- e. Elements and compounds are called pure substances because they have a unique set of chemical and physical properties.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	<u>Na</u>	<u>+1</u>
b. Copper	<u>Cu</u>	<u>+2, +1</u>
c. Tungsten	<u>W</u>	<u>+6</u>
d. Iron	<u>Fe</u>	<u>+3, +2</u>
e. Tin	<u>Sn</u>	<u>+4, +2</u>
f. Potassium	<u>K</u>	<u>+1</u>
g. Gold	<u>Au</u>	<u>+3, +1</u>
h. Silver	<u>Ag</u>	<u>+1</u>

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a. P^{3-} I

b. H_2O M

c. O_2 M

d. Au^{3+} I

e. Ge A

f. O A

g. Ca^{2+} I

h. NH_3 M

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**Ralph M. Feather, Susan Leach
Snyder, Dale T. Hesser**



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