#### CHEMICAL THERMODYNAMICS

- In open system, system and surroundings exchange...... (a) Energy only (b) Matter only (c) Energy and matter (d) Neither energy nor matter An isothermal expansion takes place at...... (a) constant pressure (b) constant volume (c) constant temperature (d) constant temperature & pressure 3) In a closed thermos flask few cubes of ice are placed. It is example of............. (a) closed system (b) open system (c) isolated system (d) thermodynamics system In an adiabatic compression, the internal energy of the system....... (a) decreases (b) increases (c) increases or decreases (d) none 5) Identify the intensive quantities from the following........ (a) volume (b) enthalpy (c) pressure (d) none 6) Which of the following is extensive property? (a) mole (b) mass (c) volume (d) all are correct 7) The heat change in a chemical reaction at constant pressure is.......... (a) AH (b) AU  $(c)\Delta T$  $(d) \Delta V$ 8) AU is always positive when...... (a) system absorbs heat and work is done on it (b) system emits heat and work is done on it. (c) system emits heat and no work is done on it (d) system absorbs heat and work is done by it 9) If in a chemical reaction, the products have less energy than the reactants, that reaction is..... (a) a reversible reaction (b) an exothermic reaction (c) an isothermal reaction (d) an endothermic reaction 10) Asmallpiece of Nametalis dropped into water. Nabursts into flame, the reaction is: (a) neutralisation (b) reversible (c) exothermic (d) endothermic
- 11) For the reaction:  $C_3H_{8(g)} + 5O_{2(g)} \rightarrow 3CO_{2(g)} + 4H_2O_{(j)}$  at constant temperature,  $\Delta H \Delta U$  is
  - (a) +RT (b) -3RT (c) +3RT (d) -RT
- 12) When NH<sub>4</sub>Cl is dissolved in water, the solution becomes cold. The change is?

  (a) endothermic (b) exothermic (c) super cooling (d) none

# **Mcqs Of Thermodynamics With Answers**

Yijin Wang

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