

Isotopes Practice Set

- What does atomic number tell you about an element?
- What does the number used to isotopes signify?
- How can you tell isotopes apart?

For each of the following isotopes, write the number of protons, neutrons, and electrons.

	Chlorine-35	Chlorine-37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

	Nitrogen-14	Nitrogen-15
# of protons	7	7
# of neutrons	7	8
# of electrons	7	7

	Carbon-12	Carbon-13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Sulfur-32	Sulfur-34
# of protons	16	16
# of electrons	16	16
# of neutrons	16	18

Fill in the isotopes names including chemical symbol, atomic number, and mass number from the values that your partner gave and the information provided.

	Isotope: 34	Isotope: 35
# of protons	17	17
# of neutrons	17	18
# of electrons	17	17

	Mass: 32	Mass: 34
# of protons	16	16
# of neutrons	16	18
# of electrons	16	16

	Carbon: 12	Carbon: 13
# of protons	6	6
# of neutrons	6	7
# of electrons	6	6

	Chlorine: 35	Chlorine: 37
# of protons	17	17
# of neutrons	18	20
# of electrons	17	17

Isotope Practice Set

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Isotope Practice Set:

Unveiling the Magic of Words: A Review of "**Isotope Practice Set**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Isotope Practice Set**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

<https://splashdogs.com/results/Resources/Documents/Holt%20Pre%20Algebra%20Workbook%20Answer%20Key.pdf>

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