

Step 8. Periodic table trends

Electronegativity

Ionization energy

He

Atomic radius

The periodic table displays various elements with their symbols and atomic numbers. Overlaid on the table are several arrows indicating trends:

- Electronegativity:** A black arrow pointing from left to right across the top of the table.
- Ionization energy:** A green arrow pointing from left to right across the top of the table, and a green arrow pointing from bottom to top on the right side of the table.
- Atomic radius:** A red arrow pointing from top to bottom on the left side of the table, and a red arrow pointing from right to left across the bottom of the table.
- Non-metallic properties:** A blue arrow pointing from the bottom-left towards the top-right.
- Metallic properties:** A light blue arrow pointing from the top-left towards the bottom-right.

Ionization Energy

Electronegativity

Atomic radius

An arrow head shows the INCREASE of the property

Example: comparing K atom with Br atom

- Atomic radius of K is HIGHER than atomic radius of Br
- Ionization energy of K is LOWER than ionisation energy of Br
- Metallic properties of K are HIGHER than metallic properties of Br



Lab Periodic Trends Answer Key Atomic Radiua

WJ Hussar



Lab Periodic Trends Answer Key Atomic Radiua:

The Enigmatic Realm of **Lab Periodic Trends Answer Key Atomic Radiua**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Lab Periodic Trends Answer Key Atomic Radiua** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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