

Final Study Guide

Buoyancy - ability of a fluid to exert an upward force on object immersed in a fluid. To float, an object must displace an amount of water that has a mass greater than the mass of the object. This usually happens in things that are hollow or filled with air.

Viscosity - a fluid's resistance to flow or how thick, or hard to pour a fluid is.

Density - mass per unit volume. Density = $\frac{\text{mass}}{\text{volume}}$

Physical Property - a property that can be measured without affecting the substance.

Ex: color, shape, size, mass, anything that can be observed with your senses

Chemical Property - a property that tells us how the substance interacts with other substances, the substance is changed in the process.

Ex: Flammability, reactivity

Physical change - a change that does not change the identity of the substances involved.

Ex: dissolving, cutting, tearing, melting, freezing

Chemical change - A change that does change the identity of the substances involved. Bonds must be broken and new bonds formed.

Signs of chemical change: Bubbles (production of a gas), precipitate (production of a new solid), new color, new odor, change in temperature.

Ex: Combustion, oxidation

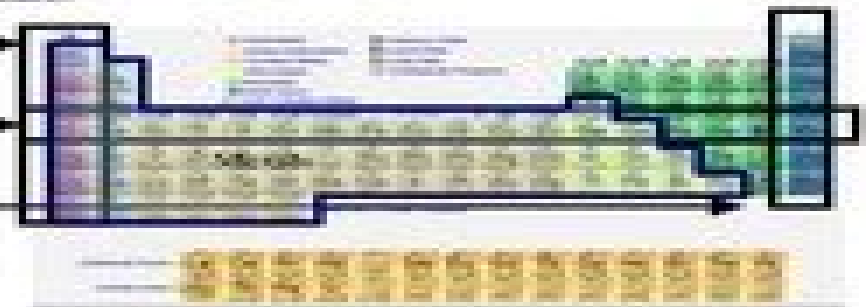
Periodic Table

Groups: the columns of the periodic table, elements in the same group have similar characteristics

Periods: the rows of the periodic table, all have

the same number of shells that an atom of a particular element have

Noble Gases - unreactive gases in group 18



symbol



Name

Atomic number: equal to the number of protons in atoms of that element

Atomic mass: weighted average mass of all the isotopes of an element

Mass number: Atomic number rounded to the nearest whole number, number of protons + number of neutrons

Metals: The elements left of the jagged line. They are usually good conductors, are malleable, are ductile, and react to magnets. All are solids except mercury.

Nonmetals: The elements to the right of the jagged line. Their properties are opposite to those of metals.



Proton: Positively charged particle found in the nucleus

Neutron: Neutral particle found in the nucleus

Electrons: Negatively charged particles found orbiting the nucleus

Glynnis Lundberg

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Table of Contents IpC Chemistry Review

1. Understanding the eBook IpC Chemistry Review
 - The Rise of Digital Reading IpC Chemistry Review
 - Advantages of eBooks Over Traditional Books
2. Identifying IpC Chemistry Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an IpC Chemistry Review
 - User-Friendly Interface
4. Exploring eBook Recommendations from IpC Chemistry Review
 - Personalized Recommendations
 - IpC Chemistry Review User Reviews and Ratings
 - IpC Chemistry Review and Bestseller Lists
5. Accessing IpC Chemistry Review Free and Paid eBooks

- IPC Chemistry Review Public Domain eBooks
- IPC Chemistry Review eBook Subscription Services
- IPC Chemistry Review Budget-Friendly Options
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 - IPC Chemistry Review Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of IPC Chemistry Review
 - Highlighting and Note-Taking IPC Chemistry Review
 - Interactive Elements IPC Chemistry Review
- 8. Staying Engaged with IPC Chemistry Review
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers IPC Chemistry Review
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection IPC Chemistry Review
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine IPC Chemistry Review
 - Setting Reading Goals IPC Chemistry Review
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of IPC Chemistry Review
 - Fact-Checking eBook Content of IPC Chemistry Review
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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