

SYLLABUS

- (i) Types of chemical changes.
 • *Direct combination* • *decomposition* • *displacement*; • *double decomposition*
 (The above to be taught with suitable chemical equations as examples).
- (ii) Energy changes in a chemical change.
Exothermic and endothermic reactions with examples – evolution/absorption of heat, light and electricity.

2.1 CHEMICAL REACTION

A chemical reaction is the process of breaking the *chemical bonds* of the reacting substances (reactants) and making new bonds to form new substances (products).

A chemical bond is the force that holds the atoms of a molecule together, as in a compound.

A chemical change or chemical reaction occurs when particles collide. Collisions occur when reactants are in close contact or by supply of energy. Thus, one or more of the following conditions are necessary for a chemical change :

(i) Mixing (close contact) : In some cases, a chemical reaction occurs when two substances are mixed in their solid states.

Example 1 : Iodine and phosphorus react explosively when brought into close contact.

2 : Lead nitrate (white) and potassium iodide (white) react to make lead iodide (yellow).



(ii) Solution : In some cases, a chemical reaction occurs when substances are mixed in, i.e. molten or aqueous state.

Example 1 : Oxalic acid crystals and sodium carbonate react in water solution only.

2 : Sodium chloride and silver nitrate also react in a solution state to form the precipitate of silver chloride and sodium nitrate.



(iii) Heat : Some chemical reactions occur only on heating.

Example 1 : Copper carbonate decomposes on heating (Δ symbol for heating) into copper oxide and carbon dioxide.



2 : Lead nitrate decomposes on heating leaving yellow residue lead monoxide, brown gas nitrogen dioxide and colourless gas oxygen.



(iv) Light : Some chemical reactions take place by the action of light. These are called *photochemical reactions* or *photolysis*. Molecules of the reactants absorb light energy to get activated, and then react rapidly.

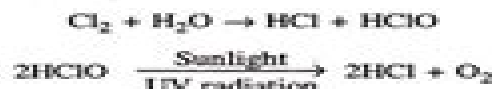
Example 1 : Plants form glucose from carbon dioxide and water in the presence of light.



2 : Hydrogen and chlorine react in the presence of sunlight.



3 : If chlorine water is exposed to sunlight, oxygen is evolved.



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