

13.2**THE NATURE OF LIQUIDS****Section Review****Objectives**

- Identify factors that determine physical properties of a liquid
- Define *evaporation* in terms of kinetic energy
- Describe the equilibrium between a liquid and its vapor
- Identify the conditions under which boiling occurs

Vocabulary

- vaporization
- evaporation
- vapor pressure
- boiling point
- normal boiling point

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Liquids are much 1 than gases. Liquids and solids are known as 2 states of matter. The conversion of a liquid to a gas or vapor is called 3. When a liquid that is not 4 changes to a gas, the process is called evaporation. A liquid evaporates faster when heated; however, evaporation itself is a 5 process. When a partially filled container of liquid is sealed, some of the particles at the 6 of the liquid vaporize. These particles collide with the walls of the container, producing a force called 7. The vapor pressure of a liquid can be determined by a device called a 8. A liquid boils when its 9 equals the external pressure. The normal boiling point of a liquid is the temperature at which the vapor pressure is equal to 10.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 11. Particles in a liquid don't have enough kinetic energy to overcome the attractive forces between them and vaporize.

Nature Of Liquids Answer Key

Scott C. Dulebohn



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Nature, 1886 *Research on the Nature of Mineral-Forming Solutions* N. P. Yermakov, 2013-10-22 Research on the Nature of Mineral Forming Solutions is the first book on the subject of fluid inclusions This book contains observational data and studies of mineral forming solutions done in the Soviet Union The description and natural classification of inclusion in minerals according to their composition and state are discussed Gaseous liquefied and solidified inclusions that are found in minerals and their significance are considered important in determining the presence and availability of the mineral For example any earlier or contemporaneous minerals that are found only in their host crystals can be determined by analyzing the presence of solid inclusions The origin and genetic classifications of liquid and gaseous inclusions being both abundant in hypogene ore deposits are explained Other less common methods in the study of inclusions besides homogenization of inclusions by heating under the microscope are forwarded The authors believe that exact measurements of the homogenization temperature are possible and therefore can serve as a precise indicator in understanding the process of formation of individual crystals and hydrothermal deposits Other studies of the All Union Research Institute of Piezo optical Mineral Raw Materials are also discussed This collection of monographs will prove invaluable to mineralogists geologists and research chemists studying minerals and ore deposits **The Nature of Matter Gr. 5-8 , Guide to RRB Non**

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