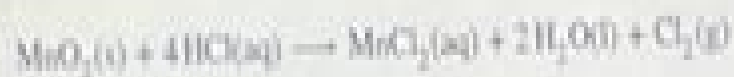


Chlorine can be prepared in the laboratory by the reaction of manganese dioxide with hydrochloric acid, HCl(aq) , as described by the chemical equation



How much $\text{MnO}_2(\text{s})$ should be added to excess HCl(aq) to obtain 115 mL $\text{Cl}_2(\text{g})$ at 25 °C and 775 Torr?

mass of MnO_2 :

13

correct

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EW Minium



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