Equilibrium Practice Problems Report

Show your work and write your answers on this page or on separate paper and submit the pictures.

R = 0.08206 Latm/K·mol = 8.3144 J/K·mol

$$K_p = K_c (RT)^{\Delta n_g}$$

1. Write the Kp expression for the reaction

 $H_2O(g) + SO_2(g) \iff H_2SO_3(g)$ (2 pts.)

Explain what it means when a reaction is at equilibrium. (3 pts.)

Write an equilibrium constant expression for the following reaction: (2pts.)
 2SO₂(g) ← O₂(g) + O₂(g)

4. Consider the reaction

$$A_2(g) + B_2(g) \iff 2AB(g)$$
.

At equilibrium there is 0.150 mol A₂, 0.105 mol B₂ and 0.000320 mol AB in a 20.0 L flask at 2300K. Calculate the equilibrium constant, K_c. (4 pts.)

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JEE Main Chapter Wise Numerical Response Questions with Solution for Chemistry (As Per NTA Latest Pattern) Career Point Kota, 2020-07-21 Whenever a student decides to prepare for any examination her his first and foremost curiosity is about the type of questions that he she has to face We feel great pleasure to present this book before you We have made an attempt to provide Chapter wise Numerical Response Questions for JEE Main as per NTA latest pattern with answer and solutions to majority of questions Solutions to the questions are not just sketch rather have been written in such a manner that the students will be able to understand the application of concept and can answer some other related questions too We firmly believe that the book in this form will definitely help a genuine hardworking student We have tried our best to keep errors out of this book Comment and criticism from readers will be highly appreciated and incorporated in the subsequent edition We wish to utilize the opportunity to place on record our special thanks to all team members of Content Development for their efforts to make this wonderful book Best Wishes Career Point Biotechology [BT] Question Bank 3000+ Questions Based on Exam Format MCQ/NAT/Written Type DIWAKAR EDUCATION HUB, 2023-09-19 IIT JAM Code BT Practice Sets 3000 Question Answer MCQ NAT written Type Highlights of Question Answer Covered All 24 Chapters of Biology Chemistry Physics Math Based MCQ NAT MSQ As Per Syllabus In Each Chapter Unit Given 125 MCO NAT Written Type In Each Unit You Will Get 125 Question Answer Based on Multiple Choice Questions MCQs Numerical Answer Type NAT Writtern Type Questions Total 3000 Questions Answer with Explanation Design by (Free Sample) 10 in One Study Package for CBSE Chemistry Class 11 with 3 Sample Professor JRF Qualified Faculties Papers Disha Experts, 10 in ONE CBSE Study Package Chemistry class 11 with 3 Sample Papers is another innovative initiative from Disha Publication This book provides the excellent approach to Master the subject The book has 10 key ingredients that will help you achieve success 1 Chapter Utility Score Evaluation of chapters on the basis of different exams 2 Exhaustive theory based on the syllabus of NCERT books 3 Concept Maps for the bird's eye view of the chapter 4 NCERT Solutions NCERT Exercise Questions 5 VSA SA LA Questions Sufficient Practice Questions divided into VSA SA LA type Numericals are also included wherever required 6 HOTS Exemplar Value Based Questions High Order Thinking Skill Based Moral Value Based and Selective NCERT Exemplar Questions included 7 Chapter Test A 15 marks test of 30 min to assess your preparation in each chapter 8 Important Formulas terms and definitions 9 Full Syllabus Sample Papers 3 papers with detailed solutions designed exactly on the latest pattern of CBSE 10 Complete Detailed Solutions of all the exercises

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