



KEY CONCEPTS

Define and explain the primary concepts and theories relevant to the subject. Include necessary terminology to ensure a solid foundational understanding.



ESSENTIAL FORMULAS

List all crucial formulas associated with the topic.
Provide a brief explanation of each formula's
purpose and context in which it is used.

PROBLEM-SOLVING TECHNIQUES

33

Outline effective methods and strategies for solving typical math problems. This might include step-by-step approaches for tackling equations, geometry problems.



PRACTICE PROBLEMS

Include a variety of problems ranging from basic to advanced levels. Provide solutions and detailed explanations to help reinforce learning.



ADDITIONAL RESOURCES

Recommend books, websites, apps, and other tools that provide further explanations, practice questions, and interactive learning experiences to enhance understanding.

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance

lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of $1 ext{ s } 2 ext{ s } 1$

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web the electron removed during the ionization of beryllium he 2s 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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