

| Question |     |      | Answer  | Mark | Guidance   |
|----------|-----|------|---|------|--|
| 2        | (a) |      | FIRST CHECK THE ANSWER ON THE ANSWER LINE<br>IF answer = $\text{CH}_4\text{N}_2\text{O}$ award 2 marks<br><br>C                      H                      N                      O<br><br>20.00/12.0    6.67/1.0            46.67/14.0    26.66/16.0<br><br>OR<br><br>1.67              6.67                  3.33              1.67    ratio of mol ✓<br><br>to give $\text{CH}_4\text{N}_2\text{O}$ ✓ | 2    | ALLOW 1.66 for C OR 1.66 for O<br><br>IGNORE Significant figures beyond the 3rd significant figure,<br>(eg ALLOW 3.3335 for N OR 1.666 for C)<br><br>ALLOW ECF from incorrectly calculated ratio of mol,<br>DO NOT ALLOW ECF from using an atomic number OR<br>any original sums inverted (eg 12.00/20.00)<br>ALLOW any order of atoms |
| 2        | (b) |      | $\text{NH}_4^+$ ✓ $\text{NO}_3^-$ ✓   | 2    | Mark incorrect ions first  |
| 2        | (c) | (i)  | $\text{H}_3\text{PO}_4$ ✓   | 1    | ALLOW formula if seen as reactant in an equation<br>IGNORE name  |
| 2        | (c) | (ii) | Calcium oxide OR calcium hydroxide OR calcium carbonate ✓   | 1    | IGNORE formulae<br>IGNORE lime, quicklime and limestone  |
| Total    |     |      |   | 6    |  |

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