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12. વર્ષ 2022-થી સંવત્સર કો-ઓપરેટિવ એગ્રીમેન્ટ હેઠળની સિખર બેઠક ક્યાં સરેરાશ યોજવામાં આવી હતી ?

(A) સમરકે

(B) નવી દિલ્લી

(C) સંવત્સર

(D) દુશાબે

13. જોડકાં જોડો.

સંધાનો પ્રકાર

સરીરના કયા ભાગમાં આવેલ છે

1. બિજાગસ સંધો

5. ખોપરીના હાડકાં વચ્ચેની સંધો

2. ખણ-દસ્તો સંધો

ખ. મૂંઠાણનો સંધો

3. ઉપાળી સંધો

ગ. નિર્તંબ તથા સ્થાપત્યના હાડકાં વચ્ચેની સંધો

4. અચળ સંધો

ઘ. ગરદન તથા માથાને જોડતી સંધો

સંધાનો પ્રકાર સરીરના કયા ભાગમાં આવેલ છે તેને જોડતાં નીચેના જવાબો પૈકી સાચો જવાબ પસંદ કરો.

(A) 1-ઘ, 2-ક, 3-ખ, 4-ગ

(B) 1-ક, 2-ખ, 3-ઘ, 4-ગ

(C) 1-ક, 2-ઘ, 3-ખ, 4-ગ

(D) 1-ખ, 2-ગ, 3-ઘ, 4-ક

14. જોડકાં જોડો.

મુળભૂત અધિકારો

અનુચેદ

1. સમાનતાનો અધિકાર

5. અનુચેદ 14

2. બંધારણીય ઉપાયોનો અધિકાર

ખ. અનુચેદ 21

3. જીવન અને વ્યક્તિગત સ્વતંત્રતાનો અધિકાર

ગ. અનુચેદ 21A

4. શિક્ષણનો અધિકાર

ઘ. અનુચેદ 32

મુળભૂત અધિકારો અને તેના અનુચેદને જોડતાં નીચેના જવાબો પૈકી સાચો જવાબ પસંદ કરો.

(A) 1-ઘ, 2-ક, 3-ખ, 4-ગ

(B) 1-ક, 2-ખ, 3-ઘ, 4-ગ

(C) 1-ક, 2-ઘ, 3-ખ, 4-ગ

(D) 1-ક, 2-ઘ, 3-ગ, 4-ખ

15. સર્વોચ્ચ અદાલતે તાજેતરમાં આવેલ ચુકાદા મુજબ ભારતના મુખ્ય મૂંઠાણી કમિશનરની નિમણૂક કોની ભલામણથી કરવાની થાય છે?

(A) મંત્રી મંડળ

(B) વડાપ્રધાન, વિરોધ પક્ષના નેતા અને કાયદા પ્રધાનની બનેલી સમિતિ

(C) વડાપ્રધાન, વિરોધ પક્ષના નેતા અને મૂંઠાણ પ્રધાનની બનેલી સમિતિ

(D) વડાપ્રધાન, વિરોધ પક્ષના નેતા અને સર્વોચ્ચ અદાલતના મુખ્ય ન્યાયાધિપતિની બનેલી સમિતિ

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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research was on the irreversible oxidation of trimethylamine by copper ii chloride which takes place at temperatures of about 75 and higher the coordination of trimethylamine by copper ii

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalise of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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