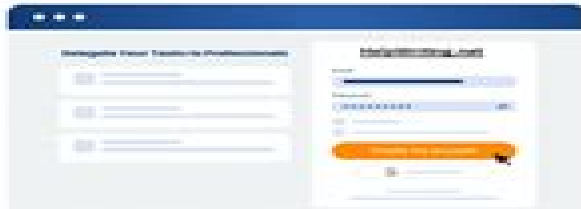
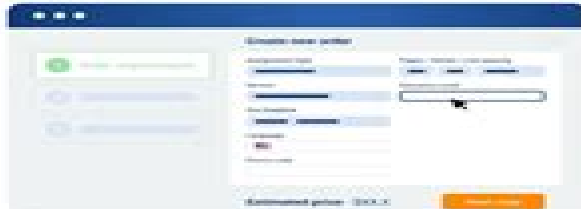


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A screenshot of the HelpWriting.net registration page. It features a sidebar with navigation links like 'Home', 'About Us', 'Contact Us', 'Privacy Policy', and 'Terms of Service'. The main content area is titled 'HelpWriting.net' and contains a registration form with fields for 'Email', 'Password', and 'Confirm Password', along with a 'Sign Up' button.

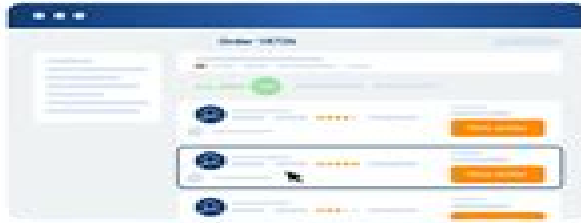
## 1. Step

To get started, you must first create an account on site **HelpWriting.net**. The registration process is quick and simple, taking just a few moments. During this process, you will need to provide a password and a valid email address.

A screenshot of the 'Create new order' page. It includes a sidebar with navigation links. The main form is titled 'Create new order' and contains fields for 'Assignment type', 'Number of pages', 'Deadline', 'Academic level', 'Subject', and 'Preferred writer'. There is also a 'Submit order' button and an 'Estimated price' of '\$200.00'.

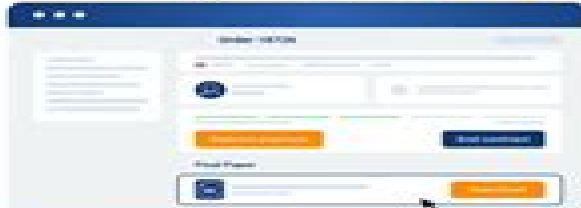
## 2. Step

In order to create a "Write My Paper For Me" request, simply complete the 10-minute order form. Provide the necessary instructions, preferred sources, and deadline. If you want the writer to imitate your writing style, attach a sample of your previous work.

A screenshot of the 'Order 18726' page. It shows a list of writer bids with columns for 'Writer', 'Rating', 'Price', and 'Status'. The first bid is from 'Writer A' with a rating of 4.5 and a price of \$200.00. There are 'View profile' and 'Place order' buttons for each bid.

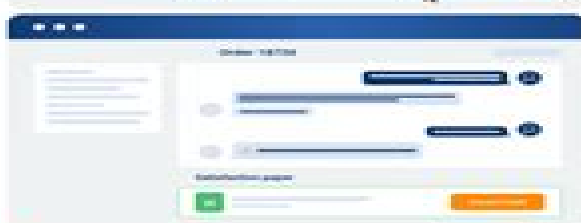
## 3. Step

When seeking assignment writing help from **HelpWriting.net**, our platform utilizes a bidding system. Review bids from our writers for your request, choose one of them based on qualifications, order history, and feedback, then place a deposit to start the assignment writing.

A screenshot of the 'Order 18726' page. It shows the 'Place order' button and a 'Total amount' of \$200.00. There is also a 'View order' button.

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 $\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$

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