

10. Two ideal gases X and Y are at the same temperature. The mass of the molecules of gas X is twice the mass of the molecules of gas Y.

What is the ratio  $\frac{\text{average speed of molecules of gas X}}{\text{average speed of molecules of gas Y}}$ ?

A.  $\frac{1}{2}$

*Microscopic level:* (absolute) **temperature** directly proportional to the **average** kinetic energy of the molecules of a substance:

B.  $\frac{1}{\sqrt{2}}$

**B**

$$\langle KE \rangle_{avg} = \frac{3}{2} kT \quad k \text{ is Boltzmann constant}$$

C.  $\sqrt{2}$

$$\frac{1}{2} m \langle v^2 \rangle_{avg} \sim T$$

D. 2

$$\frac{\frac{1}{2} m_x \langle v^2 \rangle_{x avg}}{\frac{1}{2} m_y \langle v^2 \rangle_{y avg}} = \frac{T}{T}$$

$$\frac{\langle v^2 \rangle_{x avg}}{\langle v^2 \rangle_{y avg}} = \frac{m_y}{m_x} = \frac{1}{2}$$

# M13 4 Physi Spm Tz2 Xx

**Patrick Vollmar**



**M13 4 Physi Spm Tz2 Xx:**

## Reviewing **M13 4 Physi Spm Tz2 Xx**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**M13 4 Physi Spm Tz2 Xx**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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