

# The Journal of Chemical Thermodynamics

Volume 43, Issue 12, Pages 1775-2016 (December 2011)

## Original Research Papers

**Experimental analysis and modeling of CO<sub>2</sub> solubility in AMP (2-amino-2-methyl-1-propanol) at low CO<sub>2</sub> partial pressure using the models of Deshmukh–Mather and the artificial neural network**

Pages 1775–1783

Hassan Pahlavanzadeh, Shiva Nourani, Mohammad Saber

### Highlights

► The absorption of CO<sub>2</sub> into aqueous solution of AMP at low CO<sub>2</sub> partial pressure was experimentally and analytically investigated by two different mathematical models which are Deshmukh–Mather model and artificial neural network. ► In experimental section it was observed that the solubility of CO<sub>2</sub> in AMP increases with partial pressure and decreases with temperature and concentration of solvent. ► This study indicates that the neural modeling has a better prediction of experimental CO<sub>2</sub> loadings than Deshmukh–Mather model when is compared with experimental results in this work.

**Phase equilibria study of systems composed of refined babassu oil, lauric acid, ethanol, and water at 303.2 K**

Pages 1784–1790

Érika C. D'Anton Reipert, Christianne E.C. Rodrigues, Antonio J.A. Meirelles

### Highlights

► Refining of vegetable oils. ► Deacidification of babassu oil by liquid–liquid extraction using ethanol as solvent. ► Distribution coefficient values of lauric acid and neutral oil were evaluated. ► The carbon chain length of the molecules influenced the solubility of the system.

**Vapor pressure and enthalpy of vaporization of linear aliphatic alkanediamines**

Pages 1791–1799

Vasiliy A. Pozdeev, Sergey P. Verevkin

### Highlights

► We measured vapor pressure of diamines H<sub>2</sub>N–(CH<sub>2</sub>)<sub>n</sub>–NH<sub>2</sub> with n = 3 to 12. ► Vaporization enthalpies at 298 K were derived. ► We examined consistency of new and available in the literature data. ► Enthalpies of vaporization show linear dependence on numbers n. ► Enthalpies of vaporization correlate linearly with Kovats indices.

**Thermodynamics of non-ionic surfactant Triton X-100-cationic surfactants mixtures at the cloud point**

Pages 1800–1803

Çiğdem Batıgöç, Halide Akbaş, Mesut Boz

### Highlights

► Non-ionic surfactants are used as emulsifier and solubilize in such as textile, detergent and cosmetic. ► Non-ionic surfactants occur phase separation at temperature as named the cloud point in solution. ► Dimeric surfactants

# Journal Of Chemical Thermodynamics

**N Noddings**



## **Journal Of Chemical Thermodynamics:**

## **Journal Of Chemical Thermodynamics** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Journal Of Chemical Thermodynamics**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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