

the photo. The normal vertical deflection is approximately 10 ~ 15 mm (0.39 ~ 0.59 in). If the deflection exceeds 15 mm (0.59 in) adjust the chain tension.



Adjustment

1. Loosen axle securing nut and adjuster lock nut.
2. Turn adjusting bolts left and right until the adjust marks on the chain pullers are aligned with the adjust marks on each side of the swing arm. Tighten lock nuts on adjusting bolts.



1. Axle securing nut
2. Lock nut

3. Adjusting bolt
4. Adjust mark

NOTE:

Be sure that both right and left adjusting bolts are in the corresponding positions.

3. Tighten the rear axle securing nut.

Torque: 8.5 m·kg (60 ft·lb)

4. Check brake pedal freeplay.

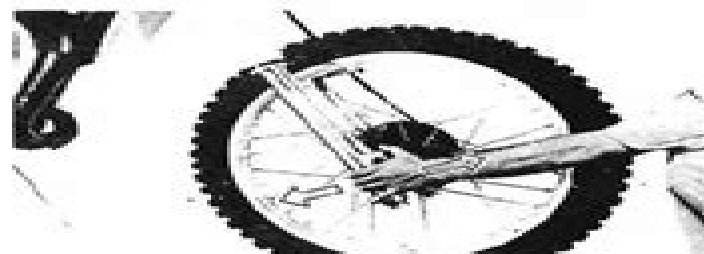
CAUTION:

Whenever the chain is adjusted and/or the rear wheel is removed, always check during reassembly:

1. Rear axle alignment.
2. Brake pedal free play.

Steering head adjustment

1. With front wheel elevated, grab bottoms of fork legs and gently push and pull to check steering head free play. There should be no noticeable free play.



NOTE:

Forks must swing from lock to lock without binding or catching. If it is felt, check the bearing or loosening.

2. To adjust, first loosen upper stem pinch bolt.
3. Loosen steering fitting bolt.

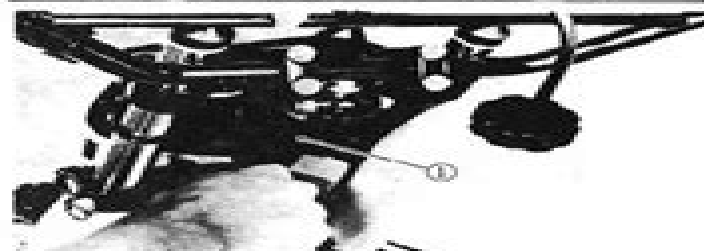


1. Steering fitting bolt

2. Stem pinch bolt

4. Use steering nut wrench to tighten ring nut. Tighten until free play is eliminated.

Tightening torque: 1.0 m·kg (7 ft·lb)



1. Steering nut wrench

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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standard molar volume figure 9 18

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