

S.L.O.P - Paper 1

Periodic table on back!

Unit 3 Quantitative Chemistry

Name:

Class:

Number of elements in a formula

Example: FeBr_3

The number of elements is the same as the number of symbols.
Each different symbol starts with a capital letter.



So we have **two** different elements **Fe** = Iron, **Br** = Bromine

Task 1: How many different **elements** are in each chemical formula?



Number of atoms of each element in a formula

Example:



The small numbers on the bottom right of each symbol tell you how many atoms of that element you have. If there is no number to the right you have one atom of that element.



Task 2: How many atoms of each element are in these formulae?
Also give the total number of atoms:



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