CHEMICAL BONDING VOCABULARY:

Chemical Bonding:

a mutual electrical attraction between the nuclei (positive) and valence electrons (negative) of different atoms that binds the atoms together.

- interaction that holds 2 atoms together.
- This forms a NEW substances with NEW properties.

Why bond? Most atoms are not stable by themselves.

A chemical bond provides **stability** for both atoms involved by lowering the potential energy of both atoms.

Stability: Atoms want 8 valence electrons for maximum stability (like a noble gas), so they either lose, gain, or share electrons from their outermost energy levels to get 8.

Exception: Helium has only 2 valence electrons in its 2nd Level which makes its outer level full & stable.

Compounds: atoms of different elements bonded together

Binary Compounds: Chemical compounds made of two distinct elements.

Two Types:

- Ionic Compounds: Compounds made of ions.
- Covalent Compounds: Chemical bond in which two atoms share valence electrons.

<u>Charge Balance</u> - The sum of positive charges equals the sum of negative charges in solution (Electroneutrality). ONLY for IONIC Compounds.

Subscripts are numbers that come after a symbol and below. Subscripts tell you the number of atoms of that element. If an element does not have a subscript, then it is understood that the subscript is 1.

A chemical nomenclature is a set of rules to generate systematic names for chemical compounds. The nomenclature used most frequently worldwide is the one created and developed by the International Union of Pure and Applied Chemistry (IUPAC).

L1 Chemical Bonding Vocab Activity

Ying Liu

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Metropolitan Detroit Science Review ,1959 Glencoe Science McGraw-Hill Staff,2001-06 Journal of Education and School World ,1886 Technical Abstract Bulletin Defense Documentation Center (U.S.),1963 University of California Union Catalog of Monographs Cataloged by the Nine Campuses from 1963 Through 1967: Authors & titles University of California (System). Institute of Library Research,University of California, Berkeley,1972 The United States Catalog Mary Burnham,Carol Hurd,1928 The United States Catalog ,1928

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 - Chapter 2: Essential Elements of L1 Chemical Bonding Vocab Activity
 - Chapter 3: L1 Chemical Bonding Vocab Activity in Everyday Life
 - Chapter 4: L1 Chemical Bonding Vocab Activity in Specific Contexts
 - ∘ Chapter 5: Conclusion
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