

# WORKSHOP MANUAL



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**A Gutmann**



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Experiment	[ $\text{CH}_3\text{COBr}$ ], $\text{mol l}^{-1}$	[ $\text{OH}^-$ ], $\text{mol l}^{-1}$	Initial rate, $\text{mol l}^{-1} \text{s}^{-1}$
1	0.10	0.10	0.025
2	0.20	0.10	0.050
3	0.10	0.20	0.050
4	0.20	0.20	0.100

a 3 10  
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potassium nitrate has an orthorhombic crystal structure at room temperature which transforms to a trigonal system at 128 °C 262 °F on cooling from 200 °C 392 °F another trigonal phase forms between 124 °C 255 °F and 100 °C 212 °F sodium nitrate is isomorphous with calcite the most stable form of calcium carbonate whereas room temperature potassium nitrate is isomorphous with aragonite a slightly less stable polymorph of

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