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### CHEMISTRY PAPER-II

#### (MODEL PAPER)

Annual Examination 2021

Total Time: 2 hours.

Total Marks: 85

Time :30 min **SECTION 'A' (M.C.Qs. (Multiple Choice Questions))** Marks: 43

Note: This section consists of 43 questions. Attempt all M.C.Qs. Each carries 1 marks.

Q 1:- Choose the correct answers for each from the given options:

The element which is present in group V A and period 3<sup>rd</sup>, its atomic number is:

- \* 15                      \* 7                      \* 8                      \* 18

2) Hydrides of group V A are ----- in nature:

- \* Acidic                      \* Basic                      \* Amphoteric                      \* Neutral

3) Potassium when combines with oxygen form:

- \* Normal oxide                      \* Super oxide                      \* Per oxide                      \* All

4) Lithium has many similarities to its diagonal neighbour-----in "Be" family:

- \* K                      \* Mg                      \* Ca                      \* Na

5)  $\text{Li}^+_{\text{aq}}/\text{Li}$  couple has exceptionally high negative electrode potential because of its large value of

- \* Ionization potential                      \* Hydration enthalpy  
\* Electron Affinity                      \* Electronegativity

6) Aluminium does not react with Nitric acid due to

- \* Low reactivity                      \* High Ionization Potential  
\* Formation of oxide layer                      \* It is a metal

7) "A" is the element of III A group which "B" belongs to V A group. When "A" reacts with "B" forms:

- \*  $\text{A}_2\text{B}_3$                       \* AB                      \*  $\text{AB}_3$                       \*  $\text{A}_3\text{B}_2$

8) The mixture of "Al" and  $\text{Fe}_2\text{O}_3$  is used in :

- \* Pyrolysis                      \* Thermite process                      \* Electrolysis                      \* Washing

9) This element is solid at room temperature and pressure:

- \* Oxygen                      \* Fluorine                      \* Bromine                      \* Iodine

10) Electronic configuration of  $\text{Cu}^{+1}$ :

- \*  $4s^1, 3d^{10}$                       \*  $4s^0, 3d^9$                       \*  $4s^0, 3d^{10}$                       \*  $4s^2, 3d^{10}$

11) The coordination number of Pt in  $[\text{Pt}(\text{en})_2\text{Cl}_2]^{+2}$  is:

# Karachi Board Chemistry For Class Xii

**Jessica J Manson**



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