

NATURAL SCIENCES

Gr. 9 - LESSON PLAN - MATTER & MATERIALS

TOPIC: COMPOUNDS		
Sub-topic: Names of Compounds		Duration: 1 Hour
CONTENT & CONCEPTS (CAPS p63)		
KEY CONCEPTS <ol style="list-style-type: none"> 1. Chemical reaction 2. Chemical formula 3. Names of compounds according to their elements. 		
Specific Aims:	Specific Aim 1: 'Doing Science'	X
	Specific Aim 2: 'Knowing the subject content and making connections'	X
	Specific Aim 3: 'Understanding the uses of Science'	
LESSON OBJECTIVES: <ol style="list-style-type: none"> 1. Learners will be able to write the names of compounds as indicated on page 63 of the CAPS Document. 2. Learners will be able to make models of different compounds 		
RESOURCES REQUIRED: <ul style="list-style-type: none"> • Periodic Table Poster • Siyavula Sasol Inzalo (Learner book) page 163 		
TEACHING & LEARNING ACTIVITY: Introduction <u>Briefly review and revise concepts dealt with in Grade 8, focusing on compounds</u> As each element has a unique name for example: <ul style="list-style-type: none"> - Hydrogen - Lithium Remember the first 20 elements learnt in grade 8. How did you memorize them? Do you remember Mnemonics... He - H; He - He; Lies - Li; Because - Be; Boys - B; Can - C; Not - N; Operate - O; Fireplaces - F; New - Ne; Nation - Na; Might - Mg; Also - Al; Sign - Si; Peace - P; Security - S; Clause - Cl A - Ar; King - K; Can - Ca LEARNERS MAY RECITE THIS MNEMONIC AS A WAY OF MEMORIZING THE FIRST 20 ELEMENTS. <u>The teacher provides the following explanation to the learners</u> Each compound has a unique name. How is a compound formed? <ul style="list-style-type: none"> - A compound is formed when TWO or more different elements combine chemically. - Each atom in a compound are held by chemical bonds - Compounds are formed during chemical reaction (when elements/ compounds combine to form new substances) - Each compound can be written as a formula, for example, H₂O; NaCl; CuSO₄ What is a formula ? <ul style="list-style-type: none"> - This is a notation that shows which elements the compound is made up of. - It indicates the total number of atoms in one unit of the compound - It indicates the ratio in which atoms of different elements are bonded to each other - It gives the total number of atoms in one unit of a compound. For example, the formula for Copper Sulphate is CuSO ₄		

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European Union over the last year Part III sets out a Southern perspective on development cooperation Part IV provides at a glance comparisons of donors aid outlook and commitment to development cooperation in the 21st Century poverty eradication gender and public support Part V contains handy reference material Throughout the book information is presented in easily interpreted diagrams and graphs The Reality of Aid has established itself as a unique source of independent evaluation and comment on aid policies and development It is indispensable for all in the field whether in the official or voluntary sectors Indispensable it gives you most of the hard facts you need to know about the major issues New Internationalist The most comprehensive and rigorous independent analysis of the aid and development policies of the world's wealthiest nations Essential reading Charity World The Reality of Aid remains an essential purchase by the libraries of development institutions and an invaluable reference for development practitioners Development Change A reliable watchdog for anyone interested in this important aspect of international relations ORBIT An accessible reference it encompasses many key issues and stimulates further research Commonwealth Comparative Politics Originally published in 1998 **Membership & NSTA Publications Catalog** National Science Teachers Association, 1997

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Table of Contents Natural Science Grade 9 Scope

1. Understanding the eBook Natural Science Grade 9 Scope
 - The Rise of Digital Reading Natural Science Grade 9 Scope
 - Advantages of eBooks Over Traditional Books
2. Identifying Natural Science Grade 9 Scope
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Natural Science Grade 9 Scope
 - User-Friendly Interface
4. Exploring eBook Recommendations from Natural Science Grade 9 Scope
 - Personalized Recommendations
 - Natural Science Grade 9 Scope User Reviews and Ratings
 - Natural Science Grade 9 Scope and Bestseller Lists
5. Accessing Natural Science Grade 9 Scope Free and Paid eBooks
 - Natural Science Grade 9 Scope Public Domain eBooks
 - Natural Science Grade 9 Scope eBook Subscription Services
 - Natural Science Grade 9 Scope Budget-Friendly Options
6. Navigating Natural Science Grade 9 Scope eBook Formats
 - ePub, PDF, MOBI, and More
 - Natural Science Grade 9 Scope Compatibility with Devices
 - Natural Science Grade 9 Scope Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Natural Science Grade 9 Scope
 - Highlighting and Note-Taking Natural Science Grade 9 Scope
 - Interactive Elements Natural Science Grade 9 Scope

8. Staying Engaged with Natural Science Grade 9 Scope
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Natural Science Grade 9 Scope
9. Balancing eBooks and Physical Books Natural Science Grade 9 Scope
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Natural Science Grade 9 Scope
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Natural Science Grade 9 Scope
 - Setting Reading Goals Natural Science Grade 9 Scope
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Natural Science Grade 9 Scope
 - Fact-Checking eBook Content of Natural Science Grade 9 Scope
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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