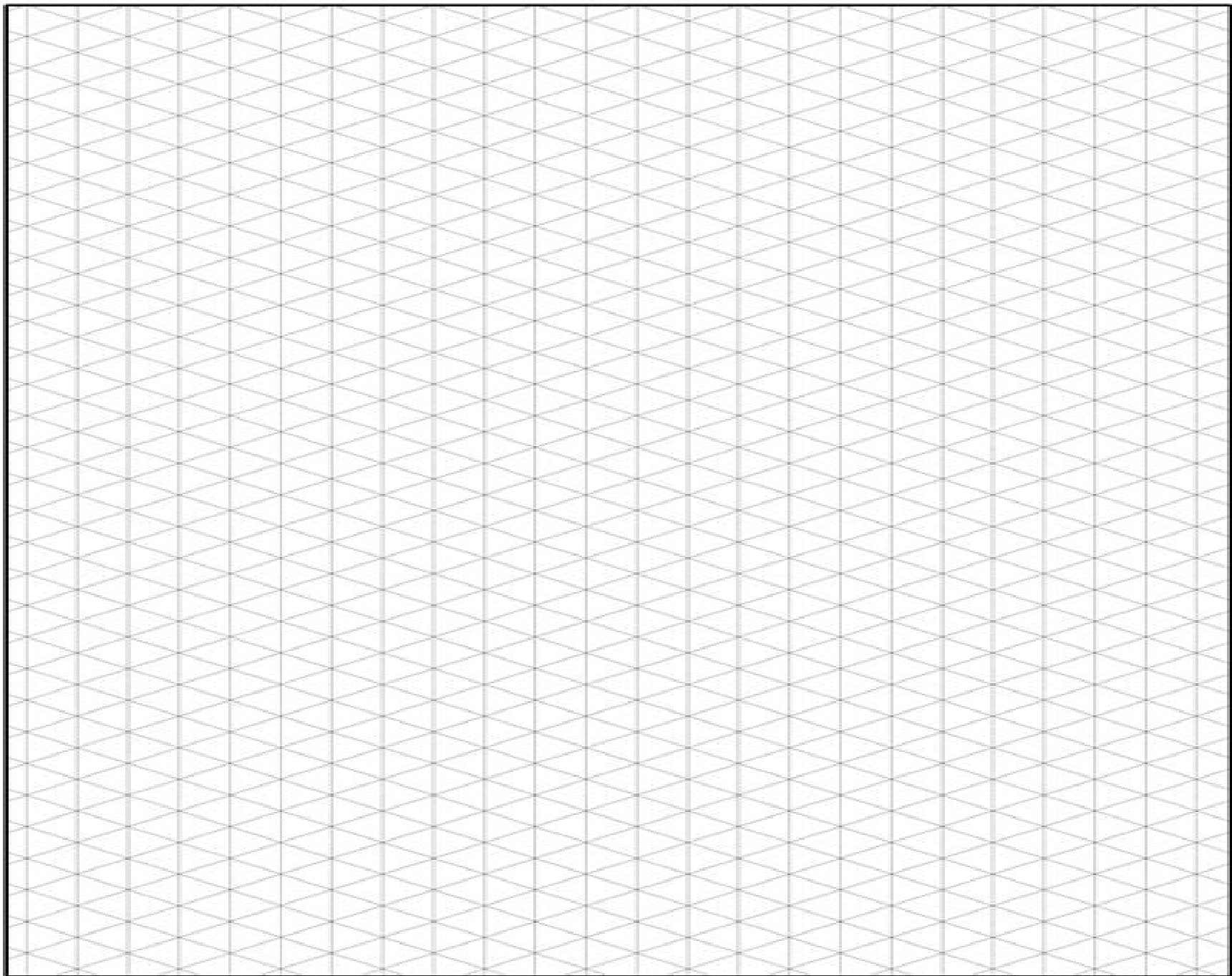




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web feb 10 2021 section 1 purpose and summary observe the color of light emitted when a sample of a salt of a metal cation is heated use the characteristic color of flames to identify ions in various laboratory samples observe the emission spectra of various elements in gas discharge tubes

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web 25 october 2022 objective experiment 1 to show how flame test can be used to identify some metals ion 2 to identify some unknown metal ions using flame test introduction the arrangement of electrons around the nucleus of the atom is

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web flame tests and atomic spectra part a flame tests of metal cations experimental data and observations solution dominant flame flame intensity flame duration wavelength nm color low medium high seconds BaCl_2 yellow to orange 597 high 13 sec CaCl_2 orange 609 high 13 sec CuCl_2 green blue medium KCl violet 423 low LiCl 701

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web oct 15 2022 the objectives of this lab exist to executing flame tested of metal cations in rank go observe their characteristic colors performance calculations to determine the frequency and energy of the emitted

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