

# Limiting and Excess Reactants

- ① For the following equation, determine which reactant is limiting and which reactant is in excess. The amount of reagents used is shown.



- ② 35.5 g of silver nitrate is reacted with 35.5 grams of sodium sulfide, producing silver sulfide and sodium nitrite.

Ⓐ Write and balance the equation.

Ⓑ Calculate the number of grams of silver sulfide produced in grams.

- ③ For the following reaction:  $\text{C}_3\text{H}_8 + 5 \text{ O}_2 \longrightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$

Ⓐ Write and balance the equation.

Ⓑ How many grams of propane is this.

# Limiting And Excess Reactants Worksheet Extension Questions

**Lifelique**

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape with a gradient, partially overlapping the bar's end.

**Limiting And Excess Reactants Worksheet Extension Questions:**

*Limiting Reactant and Percent Yield*, 2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

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