Explorationming.

Mame:		late:
Ga to poo	Student Exploration: Limiti	
	tary: chemical equation, chemical formula, chemical molecule, product, reactant, subscript	reaction, coefficient, limiting
limights y	ewledge Questions (Do these BEFORE using the you and your friends are making het dogs. A comple the store, you buy four packages of eight wieners a	bns remains a to atsismon gold tool side
1. How n	many total hot dogs can you make?	
2. Which	h ingradient limited the number of hot dogs you cou	td make?
a. Which	h ingrediere will you have leftovers of?	
now subst	Verm-up righted can be put together to make a new food, see can combine during a chamical reaction to pro- stances. The substances that undergo change are actuals. The new substances are products.	a.c.
be used a limiting re	es during a chemical reaction, one type of reactant up before the other reactants. This reactant is the reactant . Using the Limiting Reactants Glorio ¹⁷⁵ , your mine which reactant is limiting in various scenarios.	
	, make sure H ₂ + O ₂ becomes H ₂ O is selected. The B. Subscripts represent the number of atoms in a B	
1. Use to	the sliders to set the number of O_2 molecules and F	f _a melecules to two.
Α.	. How many hydrogen reciscules (H _i) are there? _	Hydrogen atoms?
В.	. How many oxygen matecules (O ₅) are there?	Oxygen atoms?
2. How n	many HyO molecules do you think will form when th	ese four maleaules react?
n. Olek F	Play (CD). How many H ₂ O molecules actually form	ned?

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developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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