

# 12.3 Limiting Reagent and Percent Yield



BY:  
MICHELLE ALICE GANIAN

## Reaction Stoichiometry

Amount of Product Formed and Percent Yield



Theoretical Yield: 0.074 g O<sub>2</sub>

Actual Yield: 0.069 g O<sub>2</sub>

Percent Yield:  $\frac{0.069 \text{ g O}_2}{0.074 \text{ g O}_2} = 93.2\% \text{ Yield}$

# Limiting Reagent And Percent Yield 12 3 Stoichiometry

**Jessica J Manson**



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