

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Student Exploration: Limiting Reactants

**Vocabulary:** chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

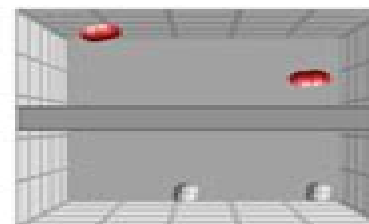
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

1. How many total hot dogs can you make?  
\_\_\_\_\_30\_\_\_\_\_
2. Which ingredient limited the number of hot dogs you could make?  
\_\_\_\_\_buns\_\_\_\_\_
3. Which ingredient will you have leftovers of?  
\_\_\_\_\_hotdogs\_\_\_\_\_

### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants Gizmo™*, you can determine which reactant is limiting in various scenarios.



To begin, make sure  $\text{H}_2 + \text{O}_2$  becomes  $\text{H}_2\text{O}$  is selected. The small "2" in  $\text{H}_2$ ,  $\text{O}_2$ , and  $\text{H}_2\text{O}$  is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

1. Use the sliders to set the number of  $\text{O}_2$  molecules and  $\text{H}_2$  molecules to two.
  - A. \_\_\_\_\_ How many hydrogen molecules ( $\text{H}_2$ ) are there? 2 Hydrogen atoms? 4
  - B. \_\_\_\_\_ How many oxygen molecules ( $\text{O}_2$ ) are there? 2 Oxygen atoms? 4



# Limiting Reactant Gizmo Answers

**Lifelique**



**Limiting Reactant Gizmo Answers:**

**Limiting Reactant and Percent** Lifeliqe,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

## **Limiting Reactant Gizmo Answers** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has are more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Limiting Reactant Gizmo Answers**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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