

## Limiting Reactants Practice



a) What is the limiting reactant if 0.866 mol of NO is mixed with 0.503 mol of O<sub>2</sub> to create NO<sub>2</sub>?

$$0.866 \text{ mol NO} \times \frac{2 \text{ mol NO}_2}{2 \text{ mol NO}} = 0.866 \text{ mol NO}_2$$

limiting reactant                      product made

$$0.503 \text{ mol O}_2 \times \frac{2 \text{ mol NO}_2}{1 \text{ mol O}_2} = 1.006 \text{ mol NO}_2$$

b) Calculate the number of moles of NO<sub>2</sub> produced. (Answer should already be solved from the previous questions)

**0.866 mol NO<sub>2</sub>**



a) If 7.40 g of O<sub>3</sub> reacts with 0.670 g of NO, which compound will be the limiting reactant?

O<sub>3</sub>  
O: 3 × 15.999 = 47.997 g/mol  
⇒ 1 mol O<sub>3</sub> = 47.997 g O<sub>3</sub>

$$7.40 \text{ g O}_3 \times \frac{1 \text{ mol O}_3}{47.997 \text{ g O}_3} \times \frac{1 \text{ mol NO}_2}{1 \text{ mol O}_3} \times \frac{46.005 \text{ g NO}_2}{1 \text{ mol NO}_2} = 7.093 \text{ g NO}_2$$

NO  
N: 1 × 14.007 = 14.007  
O: 1 × 15.999 = 15.999  
⇒ 1 mol NO = 30.006 g/mol

$$0.670 \text{ g NO} \times \frac{1 \text{ mol NO}}{30.006 \text{ g NO}} \times \frac{1 \text{ mol NO}_2}{1 \text{ mol NO}} \times \frac{46.005 \text{ g NO}_2}{1 \text{ mol NO}_2} = 1.027 \text{ g NO}_2$$

limiting reactant                      product made

b) How many grams of NO<sub>2</sub> will be produced? (Answer should already be solved from the previous questions)

**1.027 g NO<sub>2</sub>**

NO<sub>2</sub>  
N: 1 × 14.007 = 14.007  
O: 2 × 15.999 = 31.998  
⇒ 1 mol NO<sub>2</sub> = 46.005 g/mol

c) Calculate mass, in grams, of the excess reactant remaining at the end of the reaction. How many moles of excess reactant would this be?

$$0.670 \text{ g NO} \times \frac{1 \text{ mol NO}}{30.006 \text{ g NO}} \times \frac{1 \text{ mol O}_3}{1 \text{ mol NO}} \times \frac{47.997 \text{ g O}_3}{1 \text{ mol O}_3} = 1.072 \text{ g O}_3 \text{ used!}$$

$$\underbrace{7.40 \text{ g O}_3}_{\text{original}} - \underbrace{1.072 \text{ g O}_3}_{\text{used}} = \underline{6.328 \text{ g O}_3 \text{ leftover}}$$

# Limiting Reactants In A Recipe Answer Key

**Zhe Xu**



## Limiting Reactants In A Recipe Answer Key:

Foundations of Physical Science Tom Hsu PH.D.,2002      The Complete Idiot's Guide to Chemistry Ian Guch,2003 Guch covers all the elements the Periodic Table ionic and covalent compounds chemical reactions acids and bases and much more

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Prentice Hall Chemistry, 2000      **Mapping College Chemistry** Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions equilibrium acid base equilibria and electrochemistry Why would this resource be of interest to chemistry students To be successful to get into a well known college medical school physical therapy or graduate program often requires that students get an A in your pre requisite Introductory General Chemistry course To make matters worse many college professors feel that only a few students should get A grades and therefore they give difficult exams that many students fail this is the weeding out process that every pre health student is apprehensive about To succeed in this competitive environment entails not just studying harder or longer it means re organizing textbook content so that it is meaningful to the student This is the first text of its kind to employ a reliable research based strategy that incorporates a decision based visual tool to solve chemistry textbook problems ones that can make or break a career      *Meta-Analytics* Steven Simske, 2019-03-10 Meta Analytics Consensus Approaches and System Patterns for Data Analysis presents an exhaustive set of patterns for data science to use on any machine learning based data analysis task The book virtually ensures that at least one pattern will lead to better overall system behavior than the use of traditional analytics approaches The book is meta to analytics covering general analytics in sufficient detail for readers to engage with and understand hybrid or meta approaches The book has relevance to machine translation robotics biological and social sciences medical and healthcare informatics economics business and finance In addition the analytics within can be applied to predictive algorithms for everyone from police departments to sports analysts Provides comprehensive and systematic coverage of machine learning based data analysis tasks Enables rapid progress towards competency in data analysis techniques Gives exhaustive and widely applicable patterns for use by data scientists Covers hybrid or meta approaches along with general analytics Lays out information and practical guidance on data analysis for practitioners working across all sectors

**European Symposium on Computer Aided Process Engineering - 13** Andrzej Kraslawski, Ilkka Turunen, 2003-05-07 This book contains papers presented at the 13th European Symposium on Computer Aided Process Engineering ESCAPE 13 The ESCAPE symposia bring together scientists students and engineers from academia and industry who are active in the research and application of Computer Aided Process Engineering The objective of ESCAPE 13 is to promote CAPE applications into new businesses and technologies by highlighting the use of computers and information technology tools in five specific areas process design process control and dynamics modeling simulation and optimization applications in pulp

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